

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001342520017-6

PORTEVIN A.

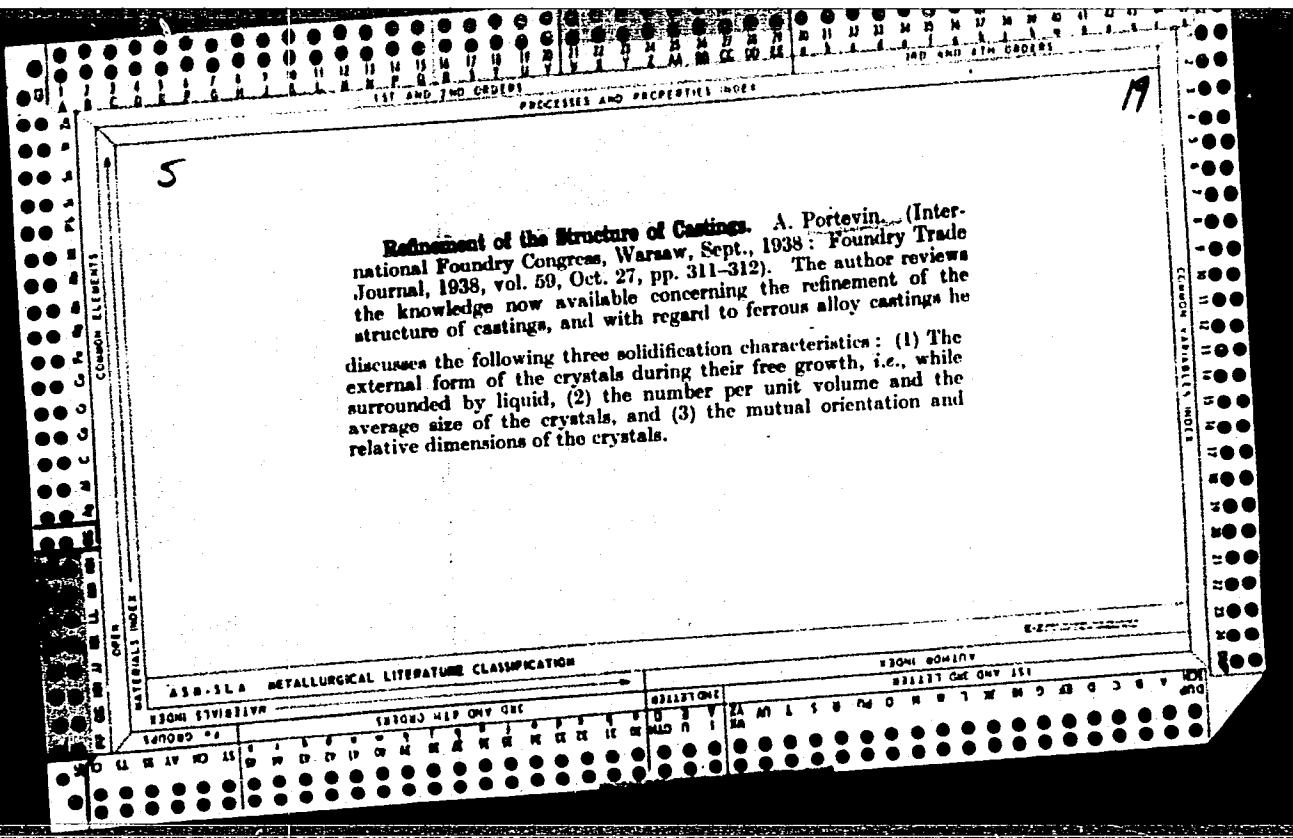
The Data and Laws of Pure Science as Sources of Practical Discoveries. A. Portevin. (*Problems and Perspectives of Educational Activities and Foundry*, 1950, 21-24). (In French and Czech). An example given is the hardening of austenite in manganese steel after quenching. ←

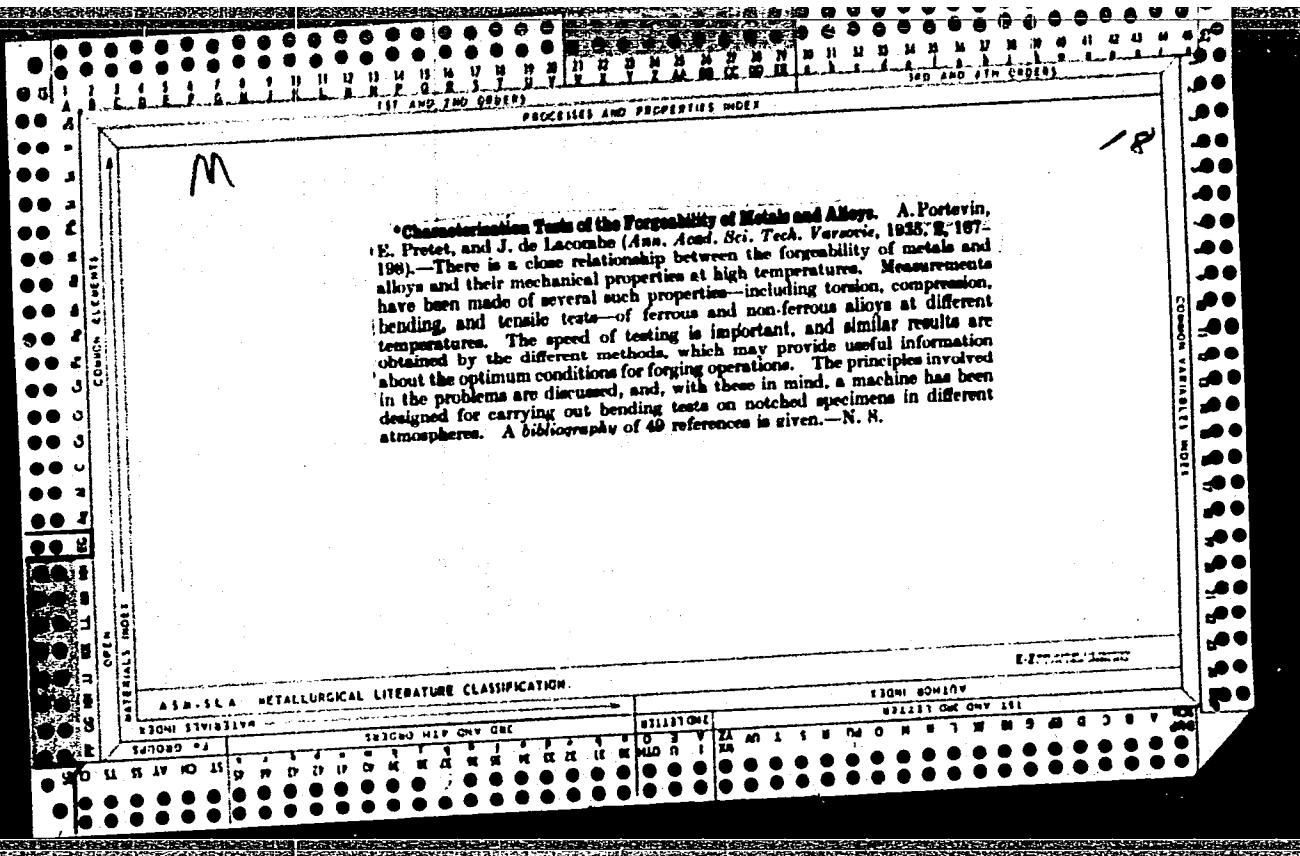
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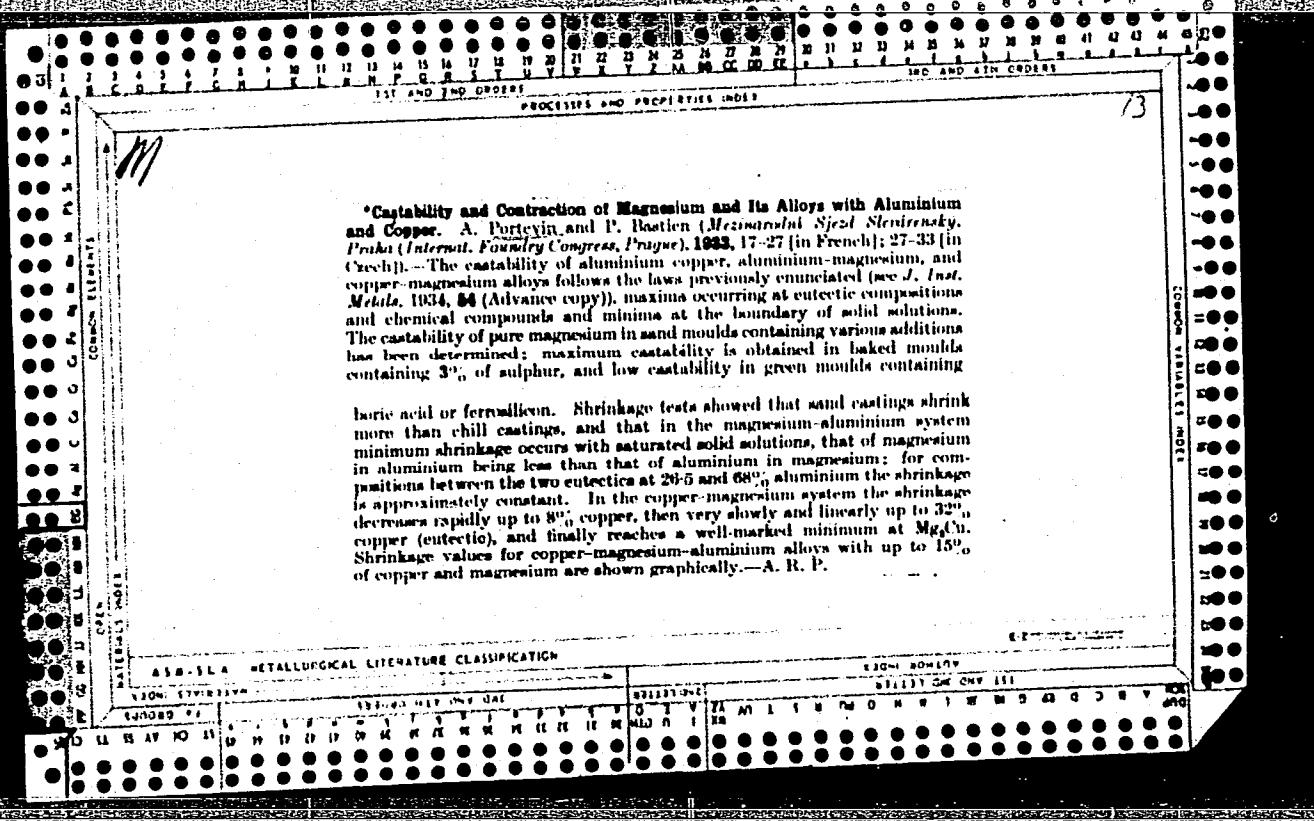
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CIA-RDP86-00513R001342520017-6"



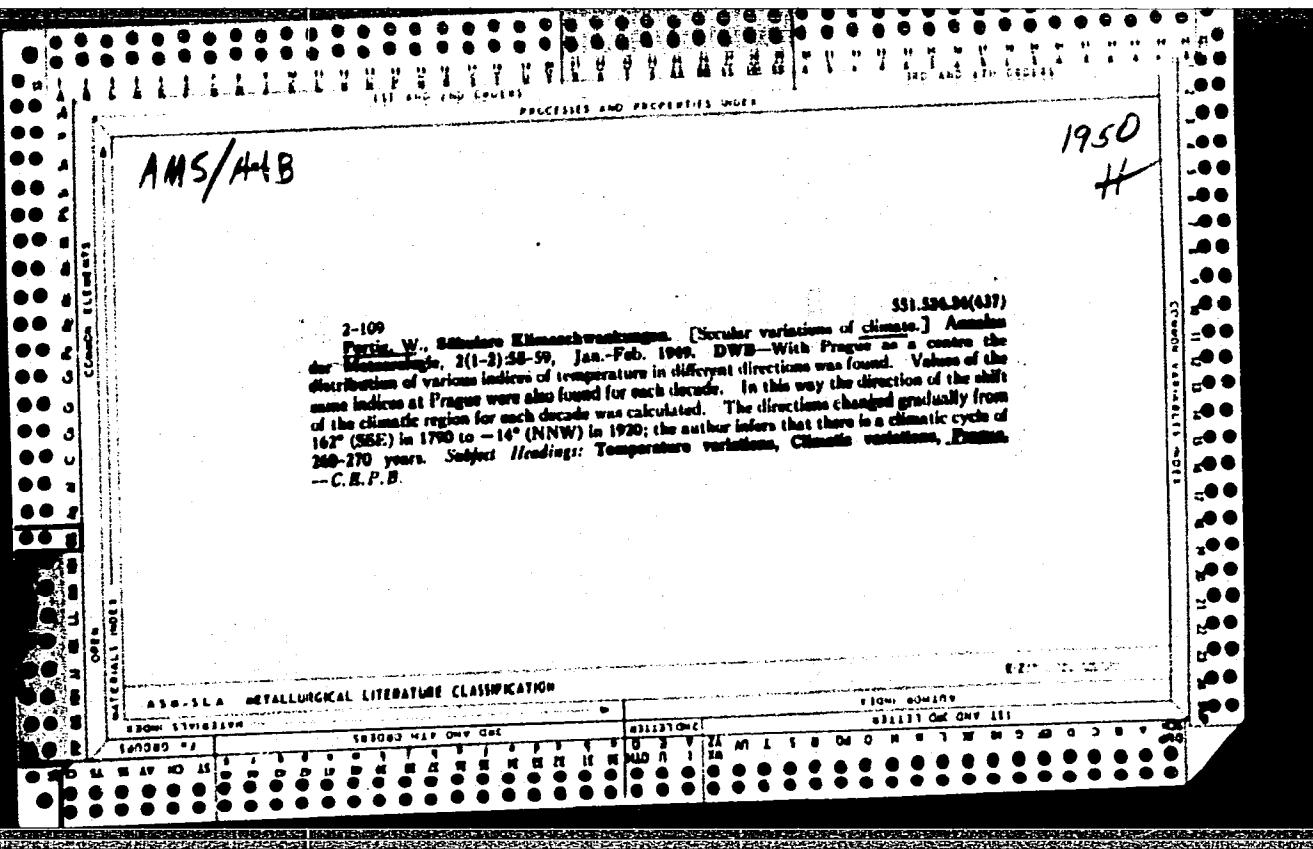


Quenching and drawing phenomena in alloys. A. Porteyin, Usine 42, No. 33, 21-4, No. 34, 24-7 (1933).—Causes and effects of quenching and tempering phenomena in steel and alloys, physicochem., structural and interior elastic modifications and the resp. states of equil. are discussed. Physicochem. equil. is defined by the equil. diagram, structural equil. is that which would be realized by the complete sepn. of phases whereby each phase would form only one cryst. individual, and stable elastic equil. exists when all internal stresses are absent. The 1st and 3rd equil. states can be obtained in practice, while the 2nd is an ideal state. All materials in a quenched state are out of equil. and each treatment which removes the equil. has a quenching effect; inversely, each treatment which brings a material closer to an equil. state has a tempering effect; this applies to all 3 kinds of equil. Methods of investigating these states (mech. tests, x-ray spectrograms, heat tests, etc.) are described, and the establishment of laws for the various conditions and possibilities of tempering is attempted. Generally valid explanations have not yet been found. Eleven references. M. Hartenstein



PORTIANKO, V.F.

✓ 16774* (Experiment in Nutrient Sprays on Grapevine.) Opyt
vnekornevoj podkormki vnegradu. V. F. Portianko. Sad i
Ogorod, 1954, no. 8, Aug., p. 61-62.
Use of superphosphate, KCl, superphosphate with B, Bordeaux
mixture with superphosphate. Table.



ACC NR: AP7006236

SOURCE CODE: UR/0076/67/041/001/0164/0169

AUTHOR: Yagodovskiy, V. D.; Portil'ya, M.

ORG: Peoples' Friendship University im. Patrice Lumumba (Universitet druzhby narodov)

TITLE: Effect of adsorption of ammonia on the resistance temperature coefficient of gold films

SOURCE: Zhurnal fizicheskoy khimii, v. 41, no. 1, 1967, 164-169

TOPIC TAGS: ammonia, adsorption, gold, metal film, work function, electric resistance

ABSTRACT: The work function ϕ of gold films (deposited on pyrex glass) before and after adsorption of ammonia was determined by measuring the electrical resistance of the films in the range of 10 to -10°C. ϕ represents the work associated with the emergence of an electron from the metal granule to the surface of the dielectric (glass). The results were treated by using the equation

$$\log (rT) = \log A + \phi/2.3 kT,$$

where A is a constant. In all cases, before and after adsorption, the dependence of $\log (rT)$ on $1/T$ was linear, indicating that the films had a granular structure. A

Card 1/2

UDC: 541.183

PORTEINVA, N. T.

PEREL'DIK, N. Sh. and PORTEINVA, N. T. "The experience in feeding silver-black foxes with small rations of animal feed," Karakul'evodstvo i zverevodstvo, 1949, No. 3, p. 44-51

SO: U-5240, 17, Dec. 53, (Letopis 'Zhurnal Statev, No. 25, 1949).

PORTOCALA, R.; DUMITRESCU, S.; IONESCU, N.I.; BRONITSKI, A.

Morphological study of strains of influenza virus isolated during
the epidemic of February-March 1959 in the Romanian People's Republic.
Stud. cercet. inframicrobiol., Bucur. 10 no.4:433-446 '59.

1. Comunicare prezentata la Institutul de inframicrobiologie al
Academiei R.P.R..
(INFLUENZA VIRUSES)

PORTNOY, I. D. and others.

Skorostnaia avtomaticheskaiia svarka pod slocem fliusa (Opyt zavoda). Sverdl'vsk, Mashgiz,
1947. 127 p.

Refers to Ural machine-building plant.

Automatic high-speed welding under a layer of flux (practice of a plant).

SO: Manufacturing and Mechanical Engineering in the Soviet Union, Library of Congress,
1953.

PORTEVA, M.

PORTEVA, M.; PLIML, J.

Synthetic analogues of the curare alkaloids. I. Quaternary salts of polybasic aliphatic ethers and thioethers. p. 36. (Collection of Czechoslovak Chemical Communications, Praha. Vol. 18, no. 6, Dec. 1953) SC: Monthly List of Best European Abstracts, (SAL), LC, Vol. 4, No. 6, June 1955, Uncl.

PORTLIK, L.; SUCHY, J.; ADAM, M.

Intensification of the smoking process. p. 399

PRUMYSL POTRAVIN. (Ministerstvo potravinarskeho prumyslu) Praha,
Czechoslovakia, Vol. 10, n^o. 8, Aug. 1952

Monthly List of East European Accessions (EEAI), LC. Vol. 9, no. 2,
Feb. 1960

Uncl.

ACC NR: AT6036599

SOURCE CODE: UR/0000/66/000/000/0234/0236

AUTHOR: Kudryashov, Yo. I.; Marennyy, A. M.; Popov, V. I.; Portman, A. I.;
Solyanov, B. I.; Sychkov, M. A.

ORG: none

TITLE: A method of irradiating biological objects on a multicharge ion
accelerator [Paper presented at the Conference on Problems of Space Medicine held
in Moscow from 24-27 May 1966]SOURCE: Konferentsiya po problemam kosmicheskoy meditsiny, 1966. Problemy
kosmicheskoy meditsiny. (Problems of space medicine); materialy konferentsii,
Moscow, 1966, 234-236TOPIC TAGS: ion accelerator, synchrocyclotron, ionizing radiation biologic effect,
radiation tissue effect

ABSTRACT:

RBE values for heavy ions can be determined from experiments with
irradiation of tissue and plant cultures. RBE determinations are necessary
for calculations of the total dose received by cosmonaut and life-support
system on a long spaceflight. At present the RBE values for heavy ions
are set at 20-30, indicating the great radiation hazard presented by galactic
rays.

Card 1/3

ACC NR: AT6036599

Experiments were conducted on the U-150 cyclotron at Dubna, which produces beams of accelerated carbon, nitrogen, and oxygen atoms with energies around 7 Mev per nucleon with currents up to 10 μ amp. Irradiation of biological objects in these direct beams is impossible, since the dose power would be measured in megarads. In order to decrease the dose power to 2-5 rad/sec, the required level for irradiation of biological objects, a special device was used. The ion flux was decreased 10^6 times by the process of scattering heavy ions on gold foil (1-2 μ thick) fixed in a vacuum chamber. The angle between beams of primary and scattered ions reached 90°. This arrangement ensured convenience of operation and the necessary uniformity of the radiation field with respect to intensity and ion energy.

A special collimator (consisting of a system of concentric sleeves) was used to further equalize the radiation field. The collimator produces some decrease in the dose power received by the biological object, which can be compensated by increasing ion currents. Irradiation monitoring was accomplished with a special ion current integrator.

During ion scattering on gold foil secondary electrons are generated, which have a spectrum with a maximum in the region of 3 kev for a 60 Mev.

Card 2/3

ACC NR: AT6036599

energy of the incident ion. A mylar film 5μ thick which divided the vacuum part of the chamber from atmospheric air was used for electron absorption. Calculations showed that the contribution of secondary radiation to the total dose absorbed by the biological object did not exceed 1%.

Since the experimental biological objects were not more than $5-10 \mu$ thick, their absorbed dose was calculated by ionization losses in polyethylene, a substance with braking ability similar to moist tissue.

[W. A. No. 22; ATD Report 66-116]

SUB CODE: 06,20 / SUBM DATE: 00May66

Card 3/3

ACC NR: AT6036519

synchrocyclotron or with Co⁶⁰ gamma rays. The dose power of protons was 35 rad/sec and of gamma rays, 3 rad/sec. The activation and luminescent methods of proton dosimetry were used. Ionization chambers were used to monitor the beam. Mitotic activity was determined immediately after gamma irradiation, and then at intervals of 12, 24, 36, and 48 hr. Similar determinations were made 10, 20, 40, and 60 hr after proton irradiation.

A definite change in mitotic activity due to gamma and proton irradiation was observed in these experiments. Immediately after gamma irradiation with all doses the mitotic index decreased, reaching 1.6-1.3 with a 1000-1500 rad dose, as compared with 5.5 in the control. With doses of gamma rays from 750 to 1500 rad the mitotic index fell to 0.5-0.6 within 12 hr. A different pattern was observed following proton irradiation: within 10 hr of irradiation with 40-450 rad the mitotic index increased approximately 50% as compared with the control. Only with large proton doses did mitotic activity decrease. Twenty hr after proton irradiation with 40-1000 rad, the mitotic index reached a low of 1.4-0.07 (1.9 in the control).

Intensive recovery of the mitotic index in the postirradiation period was

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observed with both types of radiation: the index had reached initial levels within 36-40 hr for almost all doses. Two days after gamma irradiation the mitotic index was 2-3 times higher than the initial level, whereas after proton irradiation the mitotic index recovered in three days.

Comparison of changes in mitotic activity after both proton and gamma irradiation showed the clear dose dependence of depression of mitotic activity. The same pattern of changes was observed after both types of irradiation, and quantitative relationships in observed processes were identical in both cases. [W. A. No. 22; ADD Report 66-116]

SUB CODE: 06 / SUBM DATE: 00May66

Card 3/3

L'VOVA, E.; PORTMAN, E.; SEMENOV, P.; TERKMANOV, A.; TSEYTLIN, M.;
SHAPIRO, Ya.

Pamphlet on the development of grain industry in the forth-
coming seven-year plan ("Seven-year plan for the develop-
ment of grain industry" by A.V.Borodin. Reviewed by E.L'vova
and others). Muk.-elev.prom. 25 no.9:32 S '59.
(MIRA 12:12)

1. Leningradskoye oblastnoye upravleniye khleboproduktov.
(Grain elevators) (Grain milling) (Borodin, A.V.)

PORPMANN, Georges (Bordeaux)

Pathogenesis of otospongiosis. Cesk. otolar. 5 no.1:13-19
Feb 56.

(OTOSCLEROSIS, etiol. & pathogen.
(Fr))

DASHEVSKAYA, N.M.; PORTNAYA, A.TS.; STRIZHKO, L.V.

Significance of some methods of laboratory diagnosis of epidemic hepatitis. Lab. delo no.2:87-90 '65. (MIRA 18:2)

1. Virusologicheskoye otdeleniye laboratorii (zaveduyushchiy I.I Shpits) Dnepropetrovskoy gorodskoy sanitarno-epidemiologicheskoy stantsii (glavnnyy vrach N.A. Gulyanitskiy).

10

CH

Properties of some thiazolecarboxyanines. B. S. Fort-naya, A. L. Levkoey, and N. S. Spasokukorskii (All-Union Photo Inst., Leningrad). *Doklady Akad. Nauk S.S.R.* 75, 231-3 (1950). -Introduction of either electropos. or electroneg. substituents increases the depth of color, although in the 1st case the basicity of the dye rises, and in 2nd case it falls. Absorption in EtOH soln. and basicity measurements (expressed as molality of HCl necessary in 1×10^{-4} M dye soln. in 57% EtOH to cause 50% loss of color) for the following compds. were made; no data on the prepn. are given, except the general statement of condensation of quaternary salts of the appropriate bases in pyridine with ortho-esters. Thiazolecarboxyanines with the following substituents: none in the 4,4',5,5'-positions, abs. max. 563 nm and basicity 0.6×10^{-3} with Et on the polymethine link, 4,4'-di-Me, abs. max. 556 and basicity 5.7×10^{-3} with Et, 526 and 1.2×10^{-3} with Me, abs. max. 530 with Et on the polymethine chain; 5,5'-dimethyl, 523 and 0×10^{-3} with Et, 525 and 1.1×10^{-3} with Me, 530 with Et; 4,4',5,5'-tetramethyl, 503 and 2.4×10^{-3} with H, 536 and 7×10^{-3} with Me; 4,4'-diphenyl, 550 and 5×10^{-3} with H, 530 and 1.2×10^{-3} with Me, 534 with Et; 5,5'-diphenyl, 502 and 2.1×10^{-3} with H, 566 and 6×10^{-3} with Me, 570 with Et; 4,4',5,5'-tetraphenyl, 588 and 1.8×10^{-3} with H, 562 and 1×10^{-3} with Me, 564 with Et on the polymethine link. G. M. Kosolapoff

1951

Nov. 51

PORTNAYA B. S.

USSR/ Chemistry- Photographic Sersitizers

"Reasearch in the Field of Cyanine Dyestuffs. VI. Properties of 7,7'-Bis-(Dimethylamino)-Thiacarbocyanines, " I. I. Levkoyev, E. S. Portnaya, All-Union Sci Res Cinephoto Inst

"Zhur Obshch Khim" Vol XXI, No 11, pp 2050-2055

Showed that oxidation of m-thiacetamino-dimethylaniline with K ferrocyanide in alk soln causes closing of thiazole ring in o-position with respect to dimethylamino group, forming 7-dimethylamino-2-methylbenzothiazole. Synthesized certain 7, 7'-bis-(dimethylamino)-thiacarbocyanines and studied their light absorption properties. Dimethylamino groups caused much less displacement of absorption max toward long wave lengths when introduced into 7, 7' - position than when introduced into 5, 5' - or 6, 6' -positions.

PA 194T53

PORPNAYA, B. S.

USSR/Chemistry - Photography

1 Feb 52

"The Influence of the Intramolecular Hydrogen Bond on the Color of Indoaniline Dyes Derived From Alpha-Naphthol," B. S. Portnaya, I. I. Levkoyev, N. S. Spasokukotskiy, All-Union Sci Res Cinephoto Inst

"Dok Ak Nauk SSSR" Vol LXXXII, No 4, pp 603-605

The parent indoaniline dye derived from alpha-naphthol and diethyl-p-phenylenediamine has an absorption max at 625 millimicrons. By substituting various radicals in the 1 position of the naphthol ring, the wave length is moved to the longer side in varying amounts depending on the radical. A table

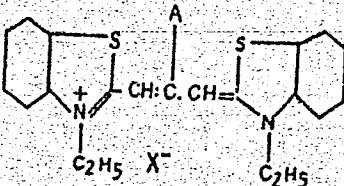
213T18

listing the radicals and the corresponding wave lengths is given. The dyes in question are used for forming the cine image in multilayer color films.

213T18

Action of Amines on Certain Thiocarbocyanines Containing an Alkoxy Alkylthio Group in the meso-Position. N. N. SVESUNIKOV, I. J. LEVKOFF, P. S. PORTNAYA and E. B. LIFSHITS. *Doklady Akad. Nauk S.S.R.*, 1952, 84:

1554/36. Thiocarbocyanines of formula:—



(X = anion and A = O.alkyl or S.alkyl) react very readily with primary and secondary amines to give a new group of thiocarbocyanines (A = NH.R, etc.), which show a hypsochromic shift, attributable to the electron donating power of the nitrogen, of 56-89 μ in their absorption maximum compared with the corresponding molecule where A = H. They react also with tertiary amines, alkyl.X- being removed with formation of symmetrical molecules containing the grouping CH.CO.CH or CH.CS.CH. For dyes in which A = NH.R, treatment with caustic soda leads to loss of HX and formation of symmetrical anils containing the grouping CH.C(NR).CH, which react with acetyl chloride to give thiacyanines for which A = NR.CO.CH₃. These show a small bathochromic shift. The maximum extinction coefficient is relatively low for all the dyes. This fact and the relatively small hypsochromic shift in the non-acylated compounds are attributed to lack of planarity in the molecules arising from steric factors.

J. Soc. Dyers and Colourists.

MF

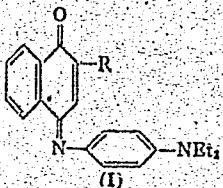
1. N. N. SVESHNIKOV, I. I. LEVKOVICH, A. F. VONPE, B. S. PORTNAYA
2. USSR (600)
4. Carbon Compounds
7. Products of reaction of acylmethylenes derivatives of N-substituted heterocyclic radicals with alkylating agents and their reactions. Dokl. AN SSSR 88 no. 2. 1953.
9. Monthly List of Russian Accessions, Library of Congress, April 1953, Uncl.

PORTNAYA, B.S.

The effect of ionization of carboxylic or sulfonic acid group in indanthrene dyes on their color. N. S. Smoskukovskii, I. I. L'vovskii, and B. S. Portnaya (All-Union Sci. Research Cine-Photo Inst., ~~USSR~~, Doklady Akad. Nauk S.S.R. 93, 671-4 (1953). — The following absorption max.

(λ_{max}) of I were observed (R and max. in acidic medium and in basic medium given): H_2 625, 625; CO_2H , 737, 600; CO_2Me , 660, 660; $CONHPh$, 709, 700; $CONHC_6H_4CO_2H$ -o, 600, 675; $CONHC_6H_4CO_2Me$ -o, 690, 690; $CONHC_6H_4CO_2H$ -m, 703, 693; $CONHC_6H_4CO_2Me$ -m, 704, 704; $CONHC_6H_4CO_2H$ -p, 710, 702; $CONHC_6H_4CO_2Et$ -p, 710, 710; SO_3H , 620, 620; $CONHC_6H_4(CO_2H)_2$ -3,5, 710, 601; $CONHC_6H_4(CO_2Me)_2$ -3,5, 710, 710; $CONHC_6H_4(CO_2H)_2$ -3,4, 712, 607;

$CONHC_6H_4(CO_2Me)_2$ -3,4, 712, 712. Thus ionization of CO_2H weakens its action as an electroneg. group; in the above dyes except for the *p*-carboxyanilide, it acts as an electropos. group giving a hypsochromic shift of absorption.



Consideration of properties of carboxylic derivs. indicates that interaction with the rest of the org. molecule takes place not only through the π -electrons, but also through the σ -electrons. The sulfo group failed to affect the absorption max. because of its complete ionization even in acid soln.

G. M. Kosolapoff

PORTNAYA, B.S.

The Committee on Stalin Prizes (of the Council of Ministers USSR) in the fields of science and inventions announces that the following scientific works, popular scientific books, and textbooks have been submitted for competition for Stalin Prize for the years 1952 and 1953. (Sovetskaya Kultura, Moscow, No. 22-40, 20 Feb - 3 Apr 1954)

<u>Name</u>	<u>Title of Work</u>	<u>Nominated by</u>
Levkoyev, I.I.	"Investigations in the	Ministry of Culture USSR
Sveshnikov, N.N.	Field of Polymethine Dyes"	
Vompe, A.F.		
Portnaya, B.S.		
Spasokukotskiy, M.S.		
Deychmeyster, N.V.		

SO: W-30604, 7 July 1954

PORINAYA, B. S., and LEVKOYEV, I. I. et. al.

"On the Interdependence of Color and Structure of Some Dyestuffs Formed in Color Development," paper given at the International Conference on Scientific Photography, Cologne, 24-27, Sep 1956.

E-3,068,138

PORTNAYA, B-S.

Azomethine dyes. I. Color of some Indoaniline dyes, derivatives of 1-naphthol, containing substituents in the naphthalene ring. B. S. Portnaya, N. S. Spasokuzotskii, N. F. Turitsyna, T. P. Bobkov, G. I. Arbuzov, and I. I. Levkoev (All-Union Cluephoto Inst., Leningrad), *Zhur. Obrabotki Khim.*, 26, 2537-40 (1958); cf. *C.A.* 49, 10504. *Chem.* b
Introduction of electroneg. groups in position 2 of Naphthol blue causes deepening of color with bathochromic shift paralleling the electronegativity of the substituent. Indoaniline dyes derived from 1-naphthol contg. CO₂H or carbamido group in the 2-position show especially deep color, apparently due to intramol. H bonding between the substituent and the carbonyl O of the naphthalene ring. Heating Ph-1-hydroxy-2-naphthalene carboxylate with amines to 135-70°, first at 40-50 min., then at 16-30 min., yielded corresponding amides of 1-hydroxy-2-naphthoic acid (I). Disubstituted amides and ethylanide were prep'd. from the acyl chloride and the amines in inert solvent. The following amides of 1-hydroxy-2-naphthoic acid were reported: anilide, 76%, m. 154°; 1-naphthylamide, 77.4%, m. 162-3°; 2-naphthylamide, 89.5%, m. 181°; amide, 101-2°, 99.2%; ethylanide, 32.5%, m. 152°; diethylamide, 25.7%, m. 67-8°; methylamide, 61.7%, m. 130°; diphenylamide, 79.6%, m. 154°; phenyl(1-naphthyl)amide, 83.1%, m. 161-2°; phenyl(2-naphthyl)amide, 87.8%, m. 146-7°. Heating 5.24 g. I-K salt in CHCl₃ with 5.2 g. PCl₅, 1.5 hrs. gave 50% pure 1-naphthol-2-sulfonyl chloride, m. 112-13° (from ligroline), which with PhNH₂ in C₆H₆ gave 70.5% 1-naphthol-2-sulfonamide, m. 148-9° (from EtOH). The use of PhNH-Et gave 91.7% N-ethyl-1-naphthol-2-sulfonamide, m. 103° (from EtOH); reaction of the chloride with Ph₂NH in 1/4

Polyakov, P. S., Spacek K. A. 1971

Et_2O in the presence of PhNMe_2 as HCl acceptor, gave 80% N,N -diphenyl-1-naphthol-2-sulfonamide, m. 135° (from Et_2O). An AgCl suspension from 7.17 g. AgNO_3 in 40 ml. H_2O and 2.92 g. NaCl in 40 ml. H_2O was treated in order with 3.2 g. dry Na_2CO_3 in 20 ml. H_2O , 0.72 g. 1-naphthol in 6 ml. EtOH ; and 1.45 g. p - $\text{Et}_2\text{NC}_6\text{H}_4\text{NH}_2$ sulfate in 40 ml. H_2O ; after stirring 0.5 hr. the mixt. was filtered, washed with H_2O , dried, extd. with C_6H_6 , the ext. was passed over Al_2O_3 and eluted with C_6H_6 , yielding at first orange by-products, then Naphthol blue; the latter parts of the eluate on evapn. gave 0.54 g. pure Naphthol blue, m. 119-20° (from MeOH). The blue pigments remaining on the adsorbent were removed with EtOEt-CHCl_3 and after chromatographing on silica in C_6H_6 , there were removed by elution with C_6H_6 3 pigments: bronze colored needles, m. 168-0°, abr. max. 635 μm , and a 3rd pigment which was not purified. Much blue-black material was still left on the column. Evapn. of the C_6H_6 soln. obtained in extn. of the AgCl-Ag ppt. above gave blue crystals, m. 130-2°, with abr. max. 626 μm , of a pigment contg. 0.20% N. The Indo-

aniline dyes listed below as analogs of α - $\text{C}_6\text{H}_4\text{CO}_2\text{CR}-$

$\text{C}_6\text{H}_4\text{NC}_6\text{H}_4\text{NHC}_6\text{H}_4-p$ were prep'd. analogously to the above synthesis of Naphthol blue (R, color, m.p., and λ in μm given): Cl, bronze, 147°, 031; Br, —, 148-0°, 638; CO_2H , blue, 126°, 602; CO_2H , blue, 172°, 732; CO_2Me , blue, 105°, 658; CONH_2 , bronze, 201°, 673; CONH_2 , blue, 112°, 672; CONH_2 , bronze, 174°, 638; CONH_2 , 2/4

Портняга, Б.С., Спасокукотский, ...

bronze, 176-0°, 692; $CONMePh$, blue-gray, 124°, 640; $CONPh_2$, bronze, 200°, 645; $CONHC_6H_4-I$, red, 194-5°, 628; $CONHC_6H_4-2$, blue, 195-7°, 626; $CONPhC_6H_4-I$, blue, 221-2°, 648; $CONPhC_6H_4-2$, blue, 197-9°, 646; SO_2NPh , bronze, 204°, 678; SO_2NPheI , bronze, 188-0°, 678; SO_2NPh , black, 181-2°, 689. II. Indoniline dyes, derivatives of 1-hydroxy-2-naphthalenone. N. F. Turiyina, B. S. Portniaya, N. S. Spasokukotskii, T. P. Bolikova, G. I. Arbusov, and I. I. Levkocev. *Ibid.* 2546-54(1050).—Introduction of substituents into the amide portion of indoniline dyes derived from 1-naphthol-2-carboxanilide does not affect the color of the dyes a great deal; electropos. groups produce small hypsochromic shifts while electroneg. groups produce bathochromic shifts of somewhat greater magnitude. A nitro group in the *o*-position shows less effect than *m*- or *p*-groups. This effect may be caused by an intramol. H bond between the amide group and O of the nitro group. Heating Ph 1-hydroxy-2-naphthalenecarboxylate with amines 1-4 hrs. at 140-70° under moderate vacuum (finally at 20 mm.) gave a distillate of PhOH, and excess amine; the residue may be steam distd. and the residue heated with aq. NaOH, the ext. being ptd. by acidification, or alternatively the residue may be exid. with EtOH or with hot AcOH. Thus were obtained 1,2-HOC₆H₄L, CONHC₆H₄R (I) (R, % yield, color, and m.p., given): *o*-Me, 6%; colorless, 110-11°; *m*-Me, 90, colorless, 124-0°; *p*-Me, 93, colorless, 154-5°; *o*-NMe₂, 78, colorless, 110-11°; *m*-NMe₂, 64, colorless, 173°; *p*-NMe₂, 50, green-yellow, 174-4°; *o*-Cl, 88, colorless, 160-2°; *m*-Cl, 85, colorless, 179-80°; *p*-Cl, 88, colorless, 172-3°; *o*-NO₂, 60, yellow, 200-1°; *m*-NO₂, 74, yellow, 243°; *p*-NO₂, 74, yellow, 241°. Reduction of I (R = *o*-NO₂) (9 g.) in 90 ml. 20% NaOH at 80-90° with 42 g. Na hydrosulfite in 150 ml. H₂O, boiling 10 min.,

3/4

Portnoy, B.S., Spesokukotsky, . . .

cooling, filtering, and acidifying gave 23.5% *H*₂*N* analog, colorless, m.p. 217-18° (from *MePh*₂), with an unidentified substance, m.p. above 260°. Similarly was prep'd. I (*R* = *m*-*H*₂*N*), m. 189-00° (insol., by-product, m. 230-1°), and the *p*-isomer, m. 180° (insol., by-product, m. 270-80°). These were heated with *AcOH*-*Ac*₂*O* yielding: I (*R* = *o*-*AcNH*), m. 214-15°; the *m*-isomer, m. 221-2°; the *p*-isomer, m. 264-5°. Heating 1.35 g. *p*-*H*₂*N*₂*C*₆*H*₄*Ac* in *C*₆*H*₆ with 1.83 g. *PhNMe*₂ and 2.06 g. 1-hydroxy-2-naphthaloyl chloride 4 hrs., followed by addn. of *Na*₂*CO*₃ and steam distn., gave on acidification of the residue 80.2% II (*R* = *p*-*Ac*), m. 194-5° (from *EtOH*). A mixt. of 0.001 mole each *p*-*Et*₂*N*₂*C*₆*H*₄*NH*, and I in aq. alc. *NaOH* (cf. part I) heated with *AgCl* 0.5-1 hr. (readily decompd. products were run at room temp.), dilut. with *C*₆*H*₆, filtered, and the org. layer worked up as described in the previous abstr. yielded the following: *o*-*C*₆*H*₄*CO*₂*C*(CONHC₆*H*₄*R*)₂*CH*₂*C*(*NC*₆*H*₄*NET*₂)₂*p* (II) (*R*, % yield, m.p., and λ in mp given): *o*-*Me*, 64, 110-11, 602; *m*-*Me*, 60, 124-6°, 691; *p*-*Me*, 92, 154-5°, 691; *o*-*NMe*₂, 78, 110-11°, 697; *m*-*NMe*₂, 87, 173°, 691; *p*-*NMe*₂, 69, 173-4°, 690; *o*-*Cl*, 88, 160-2°, 690; *m*-*Cl*, 85, 179-80°, 697; *p*-*Cl*, 88, 172-3°, 696; *o*-*NO*₂, 79, 200-1°, 698; *m*-*NO*₂, 74, 242°, 698; *p*-*NO*₂, 93, 241°, 705. The following derivs. of II gave abs. max. (mp): *o*-*NH*₂, 687; *m*-*NH*₂, 601; *p*-*NH*₂, 690; *o*-*NHAc*, 692; *m*-*NHAc*, 693; *p*-*NHAc*, 695; *o*-*Ac*, 700.

G. M. Kosolapoff

4/4

PART. 1/10
TURITSYNA, N.F.; PORTNAYA, B.S.; SPASOKUKOTSKIY, N.S.; BOBKova, T.P.;
ARBUZOV, G.I. [deceased]; LEVKOYEV, I.I.

Research in azomethane dyes. Part 2. Indoaniline dyes,
derivatives of 1,2-oxynaphthoic acid anilide. Zhur. ob. khim.
26 no. 9:2546-2554 S '56. (MLRA 9:11)

1. Vsesoyuznyy nauchno-issledovatel'skiy kinofotoinstitut.
(Aniline) (Anilides)

PORPNAYA, B.S.; SOLOV'YEVA, I.A.; TURITSYNA, N.F.; LEVKOYEV, I.I.;
CHEL'TSOV, V.S.; KRASHENINNIKOVA, M.V.; BOBKOVА, T.P.;
TKACHENKO, T.G.

Characteristics of the masking color components made of
pyrazolin arylazo derivatives and anilides of 1,2-hydroxynaph-
toic acid. Usp. nauch. fot. 8:35-43 '62. (MIRA 17:7)

PORPNAYA, B.S.; LEVKOVYEV, I.I.

Azomethine dyes. Part 8: Coloration of indoaniline dyes, derivatives of 2-acylaminophenols. Zhur. org. khim. 1 no. 12:2202-2212
(MIRA 19:1)
D '65

1. Vsesoyuznyy nauchno-issledovatel'skiy kinofotoinstitut.
Submitted December 2, 1964.

POROTNAYA, B.S.; TRACHEVSKO, T.G.; BORODINA, T.P.; CHIBUTSON, V.O.;
LEVKOYEV, I.I.

Studies in the field of azomethine dyes. Report No.7: Photographic
properties of some substituted phenols of the benzene series. Zhur.
nauch. i prikl. fot. i kin. 10 no.4:278-286 11-44 '69. (MIRA 18:7)

1. Vsesoyuznyy nauchno-issledovatel'skiy kinofotoinstitut (NIKFI).

S/058/63/000/003/045/104
A062/A101

AUTHORS: Portnaya, B. S., Solov'yeva, I. A., Turitsyna, N. F., Levkoyev, I. I., Chel'tsov, V. S., Krasheninnikova, M. V., Bobkova, T. P., Tkachenko, T. G.

TITLE: On the properties of masking color components of arylazo derived pyrazolones (5) and anilides of 1,2-oxynaphthoic acid

PERIODICAL: Referativnyy zhurnal, Fizika, no. 3, 1963, 86, abstract 3D584 ("Uspekhi nauchn. fotogr.", 1962, v. 8, 35 - 43)

TEXT: An investigation was made on the dependence of the color photographic properties of some arylazo derived pyrazolones and anilides of 1,2-oxynaphthoic acid on the nature and position of the substitution agents in the arylazo-group. It is established that the phenyl derivatives of pyrazolones and of 1,2-oxynaphthoic acid are compounds considerably less susceptible of reaction in the conditions of color developing than the initial purple and pale blue components. The entry of electropositive substitution agents into the phenylazo-group somewhat increases the reaction capacity of the components, the most favorable influence

Card 1/2

On the properties of masking color components...

3/058/63/000/003/045/104

A062/A101

then being shown by the oxy-group in the position 4. Electronegative substitution agents in the phenylazo-group of masking pale blue components cause a sharp decrease of the activity, and in the case of derivatives of 3-alkylpyrazolone they may show also a favorable influence. Some of the obtained compounds may be employed for preparing negative and contratype masking color motion-picture materials. It is shown that arylazo-derivatives of 3-alkyl- and 3-acylamino-pyrazolone usually absorb the light of the blue-violet range (maximum of absorption 400 - 420 μ). The entry of strong electron donor substitution agents into the phenylazo-group causes an appreciable deepening of their coloration. The absorption spectra of the masking pale blue components of the derivatives of 1,2-oxynaphthoic acid include the blue-violet and partially the green portion of the spectrum and in many cases they consist of two bands whose relative intensity may change strongly according to the nature and position of the substitution agents in the arylazo-group. A particularly sharp increase of the absorption intensity in the blue-violet range takes place in the case of 2-methyl- and 2-chlorophenylazo derivatives. It is established that the majority of the investigated masking purple and pale blue components at pH 5 are, as a rule, stable enough in respect to solutions containing ferrocyanic potassium. In alkaline bleaching solutions their stability strongly decreases.

[Abstracter's note: Complete translation]

Card 2/2

PORPNAYA, B.S.; BOBKOVА, T.P.; KRASHENINNIKOVA, M.V.; CHEL'TSOV, V.S.;
LEVKOYEV, I.I.

Studies in the field of azomethine dyes. Part 4: Indoaniline dyes
derivatives of 1,2-hydroxynaphthoic acid anides containing hetero-
cyclic residues in the presence of nitrogen amide. Trudy NIIFI no.
40:106-118 '60. (MIRA 15:2)

(Indoaniline)(Dyes and dyeing)

KUZNETSOV, V.A.; SINYANSKAYA, R.I.; PORTNAYA, G.N.; VOLYNSKAYA, M.P.

Electrocapillary phenomena in Te-Ag alloys and surface tension
of these alloys in a vacuum. Izv.vys.ucheb.zav.;khim.i khim.tekh.
5 no.3:428-432 '62. (MIRA 15:7)

1. Ural'skiy gosudarstvennyy universitet imeni A.M. Gor'kogo,
kafedra fizicheskoy khimii.

(Tellurium-silver alloys)

(Surface tension)

(Electrocapillary phenomena)

PORTYANKO, A.F., Cand Agr Sci -- (diss) "Cultivation of ~~wart~~
~~(the forest)~~ steppe)
birch seedlings under conditions of Omskaya Oblast." Omsk, 1959,
19 pp (Author's Report of Dissertation ~~Presented at~~ submitted to
Inst im S.M. Kirov) 120 copies (KL, 33-59, 119)

Portnaya, B.C.

USSR/Organic Chemistry. Synthetic Organic Chemistry.

E-2

Abs Jour: Ref Zhur-Khimiya, No 6, 1957, 19161

Author : Portnaya B.S., Spasokukotskiy N.S., Turitzina N.F.,
Bobkova T.P., Arbuzov G. I., Levkoyev I.I.

Inst :

Title : Studies in the Series of Azomethene Dyes. I. On the Dyeing of some Indoaniline Dyes, Derivatives of α -Naphthole, Containing Substitutes in the Naphthalene Nucleus.

Orig Pub: Zh. obshch. khimiyi, 1956, 26, No 9, 2537-2546

Abstract: Synthesis in the series of indoaniline dyes (I) is carried out by oxidation of a mixture of diethyl-n-phenylenediamine (II) and α -naphthole (III) or its derivatives, and their absorption spectra in CH₃OH is studied. To an aqueous suspension of AgCl (from 0.044 mole AgNO₃ and 0.05 mole NaCl) are added an aqueous solution of 0.03 mole Na₂CO₃, an alcoholic solution of 0.005 mole III, and an

Card : 1/5

USSR/Organic Chemistry. Synthetic Organic Chemistry.

E-2

Abs Jour: Ref Zhur-Khimiya, No 6, 1957, 19161

amines with phenyl esters IV at lowered pressure. Ethyl-amido and disubstituted amides IV are synthesized by the interaction of amines with chloranhydride IV in an inert solvent. The following amides IV are obtained (enumerated are X in COX-- substitute, yield in percent, m.p. in °C (from alc.)): C₆H₅NH, 76, 154; α-C₁₀H₇NH, 77, 4, 162-163; β-C₁₀H₇NH, 89.5, 181; NH₂, 99.2, 191-192; C₂H₅NH, 32.5, 152; (C₂H₅)₂N, 25.7, 57-58; C₆H₅(CH₃)N, 61.7, 136; (C₆H₅)₂N, 79.6, 154; C₆H₅(α-C₁₀H₇)N, 83.1, 161-162; C₆H₅(β-C₁₀H₇)N, 87.8, 146-147. 1-naphtholsulfochloride-2 (from K-salt acid and PCl₅ in CHCl₃, yield ~50%, m.p. 112-113° (from ligroin)) by the action of amines is transformed into the corresponding sulphamides (enumerated amino, yield of amides in percent, m.p. °C (from

Card : 4/5

PORPNAYA, M.S.

Supplemental adaptation of the existing obstetric phantom.
Fel'd. i akush. 26 no.10:59-62 O '61. (MIRA 14:11)

1. Meditsinskoye uchilishche Rostova-na-Donu.
(OBSTETRICS--APPARATUS AND INSTRUMENTS)

Portnaya M. S.

PoRTNAYA, M.S.; LEMZYAKOVA, Z.P.; ZYKOVA, T.Ye. (Rostov-na-Donu)

Methodical instructions on how to use M.S.Portnaya's attachment
to a standard obstetrical phantom. Fel'd. i akush. 22 no.12:40-41
D '57. (MIRA 11:2)
(OBSTETRICS--AUDIO-VISUAL AIDS)

1964-1965

LEMZYAKOVA, Z.P.(Rostov-na-Donu); ZYKOVA, T.Ye.(Rostov-na-Donu); PORTNAYA,
M.S.(Rostov-na-Donu)

Methods for conducting preclinical practice on phantoms in an
obstetrics course. Fel'd. i akush. 22 no.1:49-63 Ja '57
(MIRA 10:4)

(OBSTETRICS--STUDY AND TEACHING)

LASHKEVICH, A.M.; TERENT'YEVA, A.A.; IVANOVA, L.S.; BOGDULINA, M.A.;
VELICHENKO, I.N.; NIKULENKO, V.S.; KONSHINA, T.I.; SHAKHOVA, T.P.;
NYASHINA, A.A.; YASINSKAYA, Z.A.; AGAL'TSEVA, N.B.; SEL'MENSKAYA,
Ye.G.; KRETSMER, V.L.; KONONOVICH, L.K.; FEDORAYEVA, A.M.; TKACHUK,
L.Ya.; VYATKINA, G.A.; SLOUSHCH, V.S.; RACHINSKAYA, L.N.; PORTNAYA,
R.Yu.; KARAKOVSKAYA, E.M.; POKROVSKAYA, M.A.; KORNEVA, A.I.;
YERSHOVA, K.F., otv. red.; Prinimal uchastiye KAMANOV, M.I., red.;
LAGAREVA, A.P., otv. za vypusk; NIKITINA, I.P., tekhn. red.

[Economy of Novosibirsk Province; collection of statistics] Narodnoe
khoziaistvo Novosibirskoi oblasti; statisticheskii sbornik. Novo-
sibirsk, Gosstatizdat TsSU SSSR, 1961. 331 p. (MIRA 15:6)

1. Novosibirsk. Oblastnoye statisticheskoye upravleniye. 2. Na-
chal'nik Statisticheskogo Upravleniya Novosibirskoy oblasti (for
Yershov). 3. Zamestitel' nachal'nika Statisticheskogo Upravleniya
Novosibirskoy oblasti (for Kamanov).

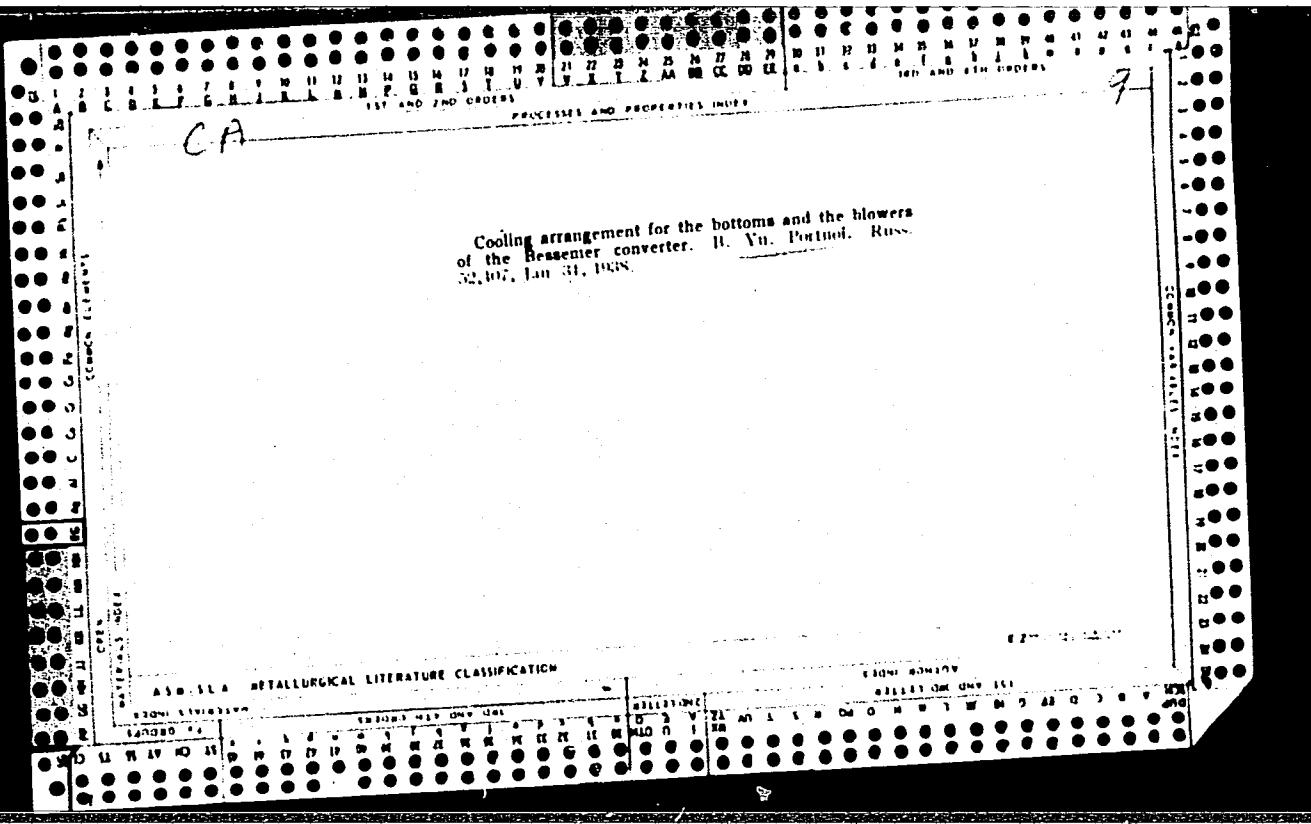
(Novosibirsk Province—Economic conditions)

ALEKSANDROVSKIY, V.A.; NIKITIN, I.; ZHITKOV, A.M.; USHMAYEV, N.;
BRYAUSHNOV, P.N.; PORTNIK, Kh.; TARLAVSKAYA, S.A.;
ALIYEV, A.A.; KEMIYA, T.

Information and brief news. Veterinariia 40 no.6:87-93
Je '63. (MIRA 17:1)

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001342520017-6

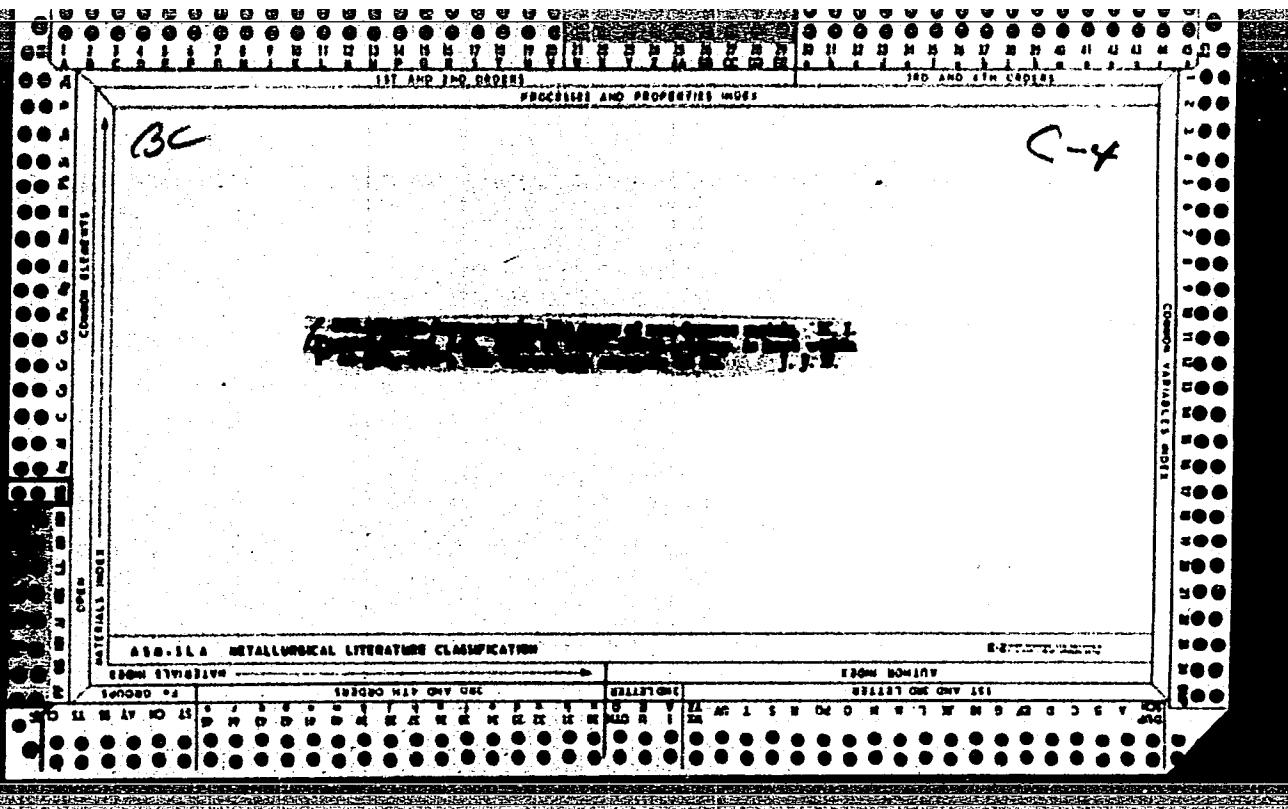


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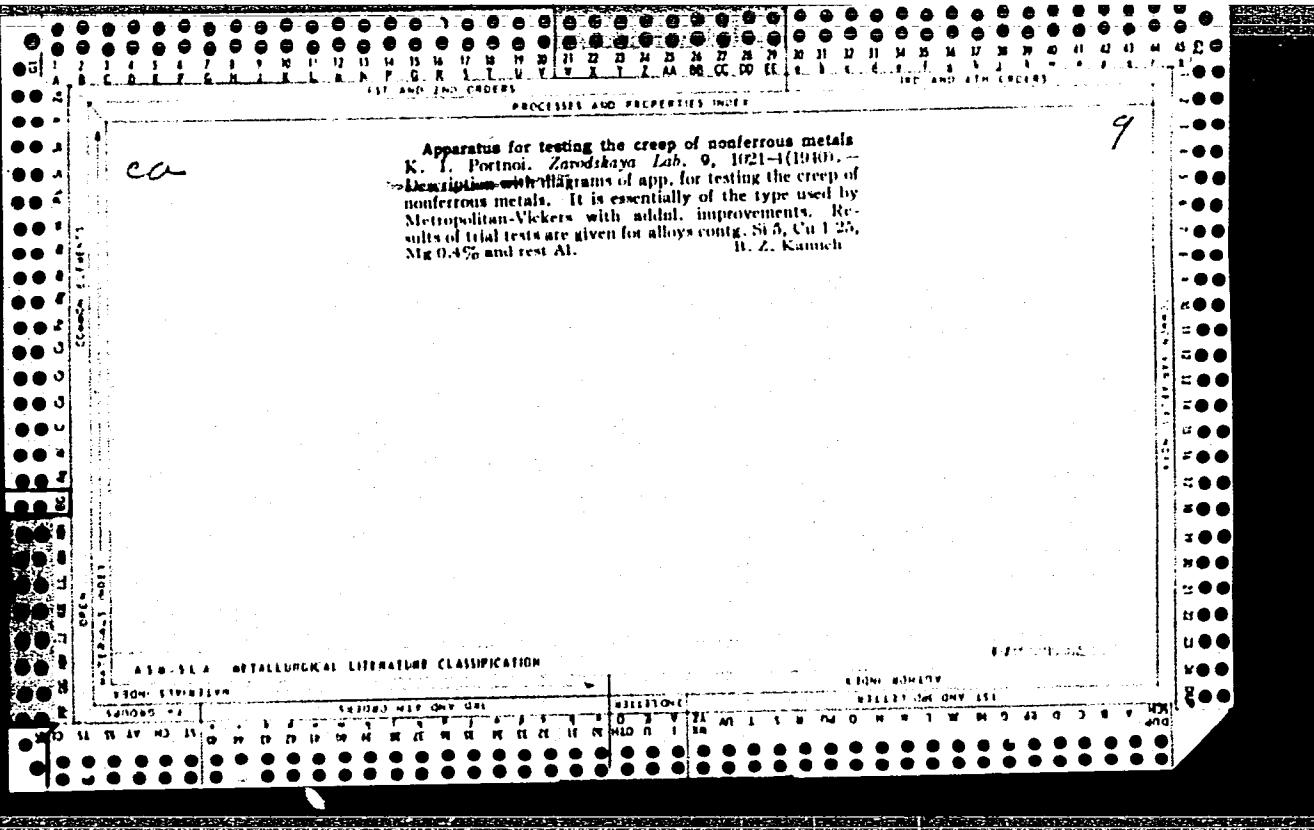
PORNOI, K. I.
A. A. BOCHVAR, IAN/OTN, 1943, n. 9/10, 3-20

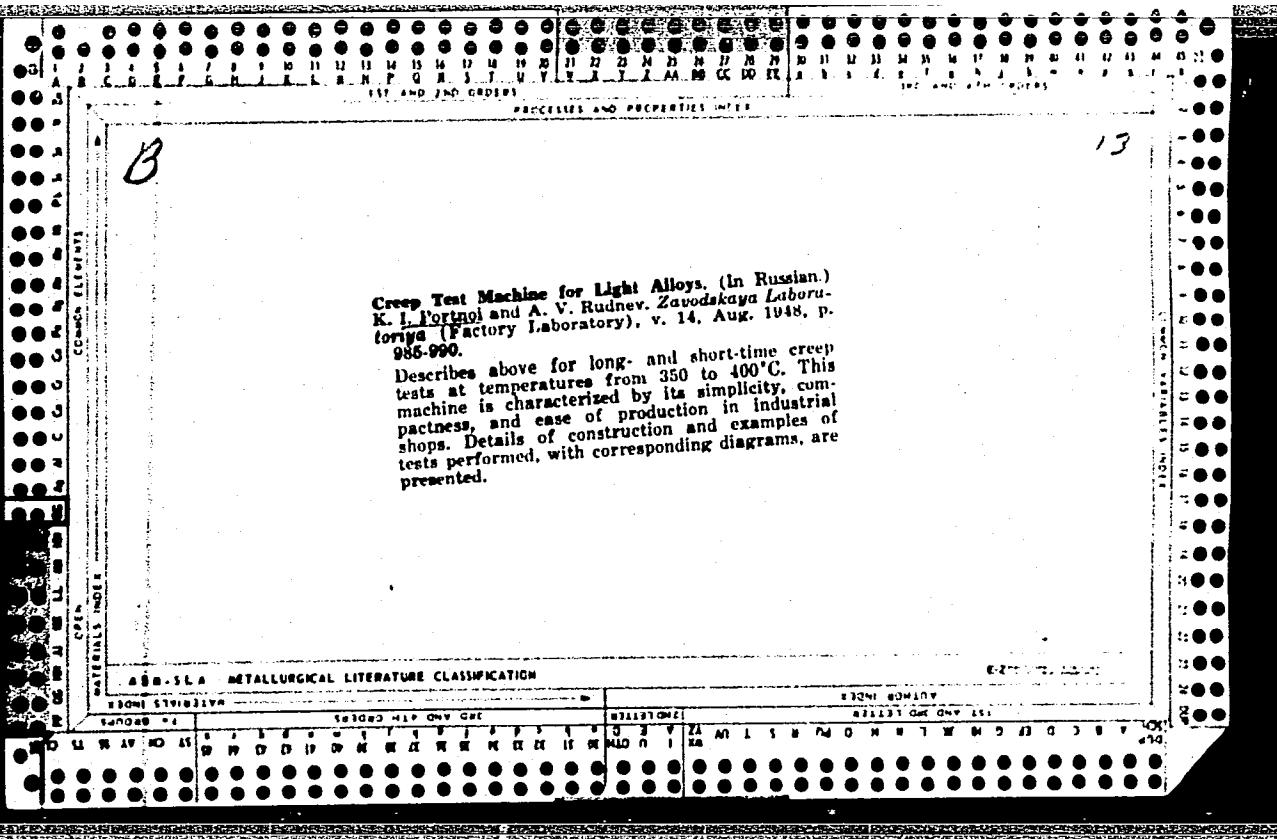
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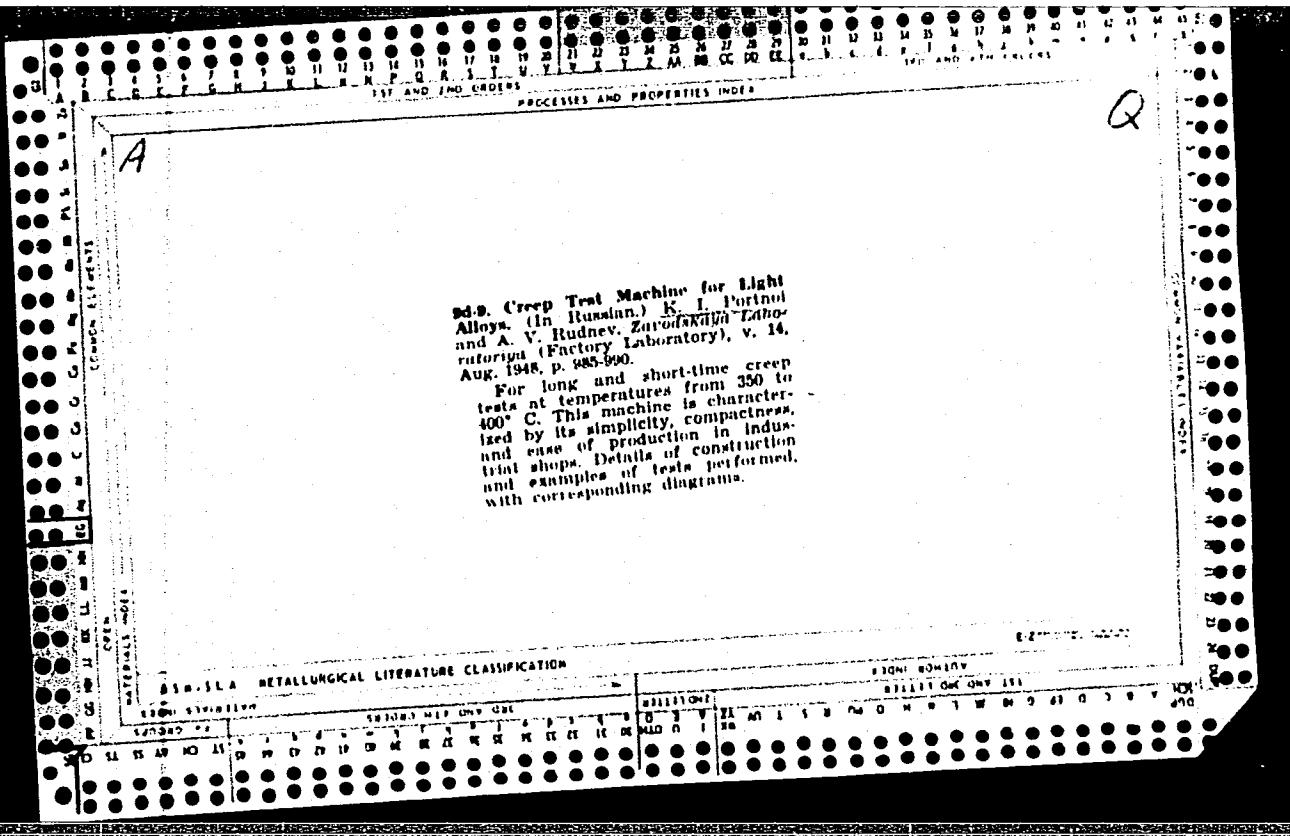
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1. PCRTNCI, K. I.
2. USSR (600)
4. Technology
7. Magnesium alloys. Moskva, Metallurgizdat, 1952.

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Fruit Culture

On the subject of a review ("Fruit and berry orchards in the central zone of the U.S.S.R." Reviewed by L. R. Portnoi), Sad i og., no. 1, 1952.

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S.A.
Sect. B

Relays - Electronic Circuits

621.318.563 : 621.313.322 : 621.316.729

772. Frequency-difference relays of the induction type for the self-synchronization of generators. M. G. PONOMARENKO AND YU. M. ELKIND. Elektrichesko, No. 9, 58-62 (Sept., 1951) In Russian.

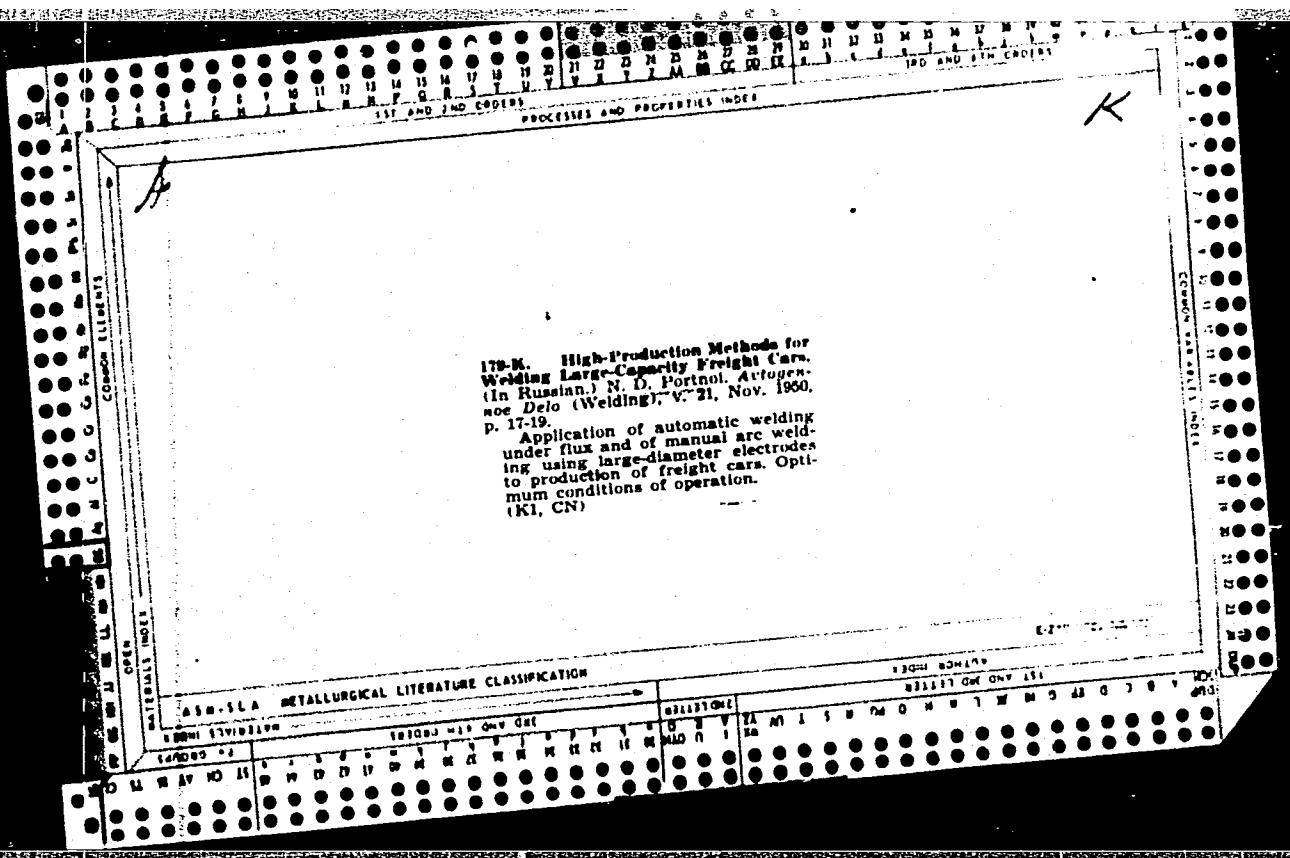
The conditions for reliable operation of this type of relay are (1) that it should give an impulse for the throwing-on of the generator when the frequency difference between the latter and the system is about 1 c/s; (2) that it should work on the residual voltage of the motor (which may be as low as 0.1 V); (3) that it should permit voltage variations of the system within the limits 0.5-1.2 x rated voltage and of variations of the frequency of the system between 40 and 52 c/s, at a maximum rate of 0.5 c/s/sec; (4) that it does not give erroneous impulses for closing the generator circuit-breaker on interruptions of the supply. The IRCH relay which satisfies these conditions is described and its theory is given. In addition, amplitude characteristics and test results of this relay are presented and the conditions for minimum influence of voltage variations in the system on the relay operation are derived. The relation between the frequency difference producing a response of the relay and the system frequency is also found.

B. F. KRAUS

PORTRDI N. D.

179-K. High-Production Methods for Welding Large-Capacity Freight Cars. (In Russian) N.D.Portnoi. Avto. Delo (Welding) v.21, Nov. 1950, p. 17-19.

Application of automatic welding under flux and of manual arc welding using large-diameter electrodes to production of freight cars. Optimum conditions of operation. (KI,CN)



PORNOV, A

A

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.P8

NEVROZY, RFAKTIVNNYE PSIKHOZY I
PSIKHOPATII [NEUROSES, PSYCHOSES, AND
PSCHOPATHY, BY] A.A. PORNOV I D.D.
FEDOTOV. MOSKVA, 1957.

123 [17] p.

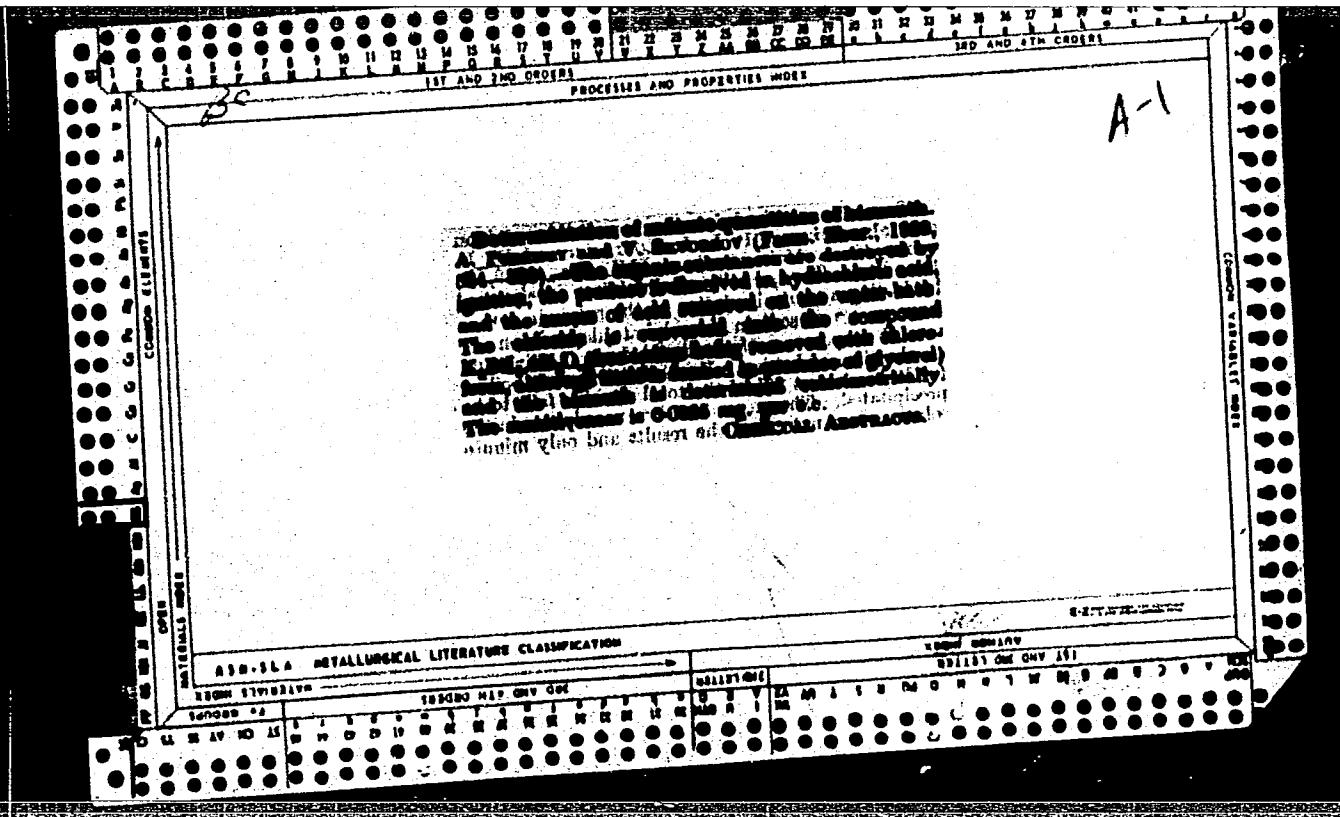
AT HEAD OF TITLE: RUSSIA. MINISTERSTVO
ZDRAVOOKHRANENIYA. [MOSCOW] NAUCHNO--
ISSLEDOVATEL'SKIY INSTITUT PSIKHIATRII.

"LITERATURA": p. 115-[124]

POROMOV, A., (Dr.), Physician

Author of article, "Harmful Remnants of the Past," concerning alcoholism.
Krasnaya Zvezda, Moscow, 12 Sep 54

SO: SUM 291, 2 Dec 1954



"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001342520017-6

PORTNOY, A. A.

Multiple drilling in coal mines of the Moscow Region Moskva Moskovskii rabochii, 1946. 35 p. (50-20797)

TN808.R92M6

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CIA-RDP86-00513R001342520017-6"

PORTNOV A.A.

KADYRBAYEV, R.A., gorny inzhener; LINNIK, G.F., gorny inzhener; PORTNOV, A.A.,
gorny inzhener.

Progressive mining practices in the Dzerzhinskii mines. Gor.zhur.no.9:
16-18 S '56. (MIRA 9:10)
(Dneprodzerzhinsk--Iron mines and mining)

PORTNOV, A.A., obshchiiy red.; BABAYAN, E.A., red.; BORINEVICH, V.V., red.; GUREVICH, Ye.I., red.; PYATNITSKAYA, I.N., red.; ROZHNOV, V.Ye., red.; STREL'CHUK, I.V., red.; FEDOTOV, D.D., red.; EHMELEV, N.S., red.

[Alcoholism; a collection of articles on its clinical aspects, pathogenesis, treatment, and prevention] Alkogolizm; sbornik rabot po klinike, patogenezhu, lecheniu i profilaktike. Pod obshchey red. A.A. Portnova. Moskva, 1959. 447 p. (MIRA 13:3)

1. Russia (1923- U.S.S.R.) Ministerstvo zdravookhraneniya.
(ALCOHOLISM)

VANS.HCHIKOV, V.M.; PORTNOV, A.A.; KHACHATURIAN, A.A.

Idealistic nature of SHmar'ian's "Brain pathology and psychiatry".
Nevropat.psikiyat., Moskva 20 no.2:70-78 Mar-Apr 1951. (CLML 20:9)

1. Docent Vanshchikov; Candidate Medical Sciences A.A. Khachaturyan.
2. Of the Institute of Psychiatry of the Ministry of Public Health USSR (Director--Prof. V.A. Gilyarovskiy).

PORNOV, A. A;REPOORT, A. M.

Planning of new psychoneurologic hospitals. Zh. nevropat.
psichiat., Moskva 52 no.5:52-58 May 1952. (CLML 22:2)

1. Moscow.

1. PORTNOV, A. A.
2. USSR (600)
4. Psychiatry
7. Certain debatable problems in psychiatry.
Zhur. nevr. i psikh. 52 №.10, 1952

9. Monthly List of Russian Accessions, Library of Congress, January 1953. Unclassified.

PORTNOV, A.A.

S. S. Korsakov Journal of Neuropathology and Psychiatry. Zh. nevropat. psichiat., Moskva 53 no.7:558-560 July 1953. (CIML 25:4)

1. Based on the materials discussed at a meeting of the branches of the All-Union Scientific Society of Neuropathologists and Psychiatrists.
2. Moscow.

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001342520017-6

PORTNOV, A.A.

The prevention of neuroses. Zdorov'e 1 no.12:28-29 D '55. (MIRA 9:2)
(NEUROSES)

APPROVED FOR RELEASE: 06/15/2000

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PORNOV, A.A.

[Clinical treatment of acute tetraencephalopathy] Klinika ostrogo
tetraenkefala psichoz. Moskva, M-vo zdravookhraneniia
SSSR, 1956. 181 p.
(PSYCHOSES) (MLRA 10:6)

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001342520017-6

PORNOV, A.A.

About drunkenness and drunks. Zdorov'e 2 no.6:20-22 Je '56.
(TEMPERANCE) (MLRA 9:8)

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001342520017-6"

ACC NR: AP7004966

SOURCE CODE: UR/0048/66/030/009/1441/1442

AUTHOR: Bogan, Ya.R.; Vitol, I.K.; Portnov, A.A.

ORG: none

TITLE: Use of the luminescent probe technique for investigating hole processes in crystal phosphors /Report, Fourteenth All-Union Conference on Luminescence (Crystal Phosphors) held at Riga, 16-23 Sept. 1965/

SOURCE: AN SSSR. Izvestiya. Seriya fizicheskaya, v. 30, no. 9, 1966, 1441-1442

TOPIC TAGS: luminescence, electron hole, exciton, F band, luminescence center, luminescent crystal, potassium bromide, thallium

ABSTRACT: The authors propose a technique, which they call "luminescent probing", for investigating hole (and other) processes in crystal phosphors whose electrical conductivities are too low to permit application of the more usual techniques involving conductivity, the Hall effect, photopolarization, etc. To employ this technique one induces in the specimen luminescence centers that radiate on interaction with only one type of mobile defect (e.g., only with holes, electrons, excitons, or the like) and investigates the luminescence and absorption in different bands. F centers radiate in the α band on interaction with V_k centers and are thus suitable centers for use in connection with luminescent probe investigations. This was confirmed by measurements of the temperature dependences of the α luminescence, the Tl

Card 1/2

ACC NR: AP7004966

luminescence, and the F absorption of a KBr:Tl crystal phosphor. Strong α luminescence, weak Tl luminescence, and a rapidly decreasing F absorption was observed in the temperature range from 165 to 180 K in which the V_k centers decay. The advantage of the luminescent probe technique is its simplicity. It is not yet known whether suitable centers can be found for study of electron and exciton processes, and the use of F centers as luminescent probes is limited by our lack of knowledge of the conditions under which α luminescence can appear as a result of exciton reactions and resonance transfer of energy. Orig. art. has: 1 figure.

SUB CODE: 20 SUBM DATE: none ORIG. REF: 010 OTH REF: 004

Card 2/2

PORNOV, A.A., kand. tekhn. nauk; KARPOV, A.A., inzh.; LEONT'YEV, A.I.,
inzh.; LEONT'YEVA, T.S., inzh.

Study of an experimental compartment furnace during the heating
of square billets. Stal' 25 no.4:370-372 Ap '65.

(MIRA 18:11)

PORNOV, Anatoliy Aleksandrovich; FEDOTOV, Dmitriy Dmitriyevich;
ROMASENKO, V.A., red.

[Psychiatry] Psikiatriia. Izd. 2., perer. i dop. Mo-
skva, Meditsina, 1965. 277 p. (MIRA 18:5)

TSIMMERMAN, R.R., inzh.; PORTNOV, A.A., glavnnyy red.; GREGORSHIKIN, I.I., zames-titel' glavnogo red.; ZELIKOV, K.N., red.; PODYIMSHCHIKOV, N.V., red.; TSITRIN, M.A., red.; STESIN, Ye.L., red.

[Calculation of mine dust removing equipment.] Raschet shakhtnykh pyleotsasyvaiushchikh ustyanovok. Moskva, Gosgortekhizdat, 1963.
62 p. (Tula. Podmoskovnyi nauchno-issledovatel'skii i proektno-konstruktorskii ugol'nyi institut. Sbornik nauchnykh trudov, no.8)
(MIRA 17:10)

FEDOTOV, D.D.; PORTNOV, A.A. (Moskva)

Clinical, social and hygienic problems of alcoholism in V.M.
Bekhterev's works. Trudy Gos. nauch.-issl. inst. psikh. 38:
427-433 '63 (MIRA 16:11)

PORNOV, Anatoliy Aleksandrovich; FEDOTOV, Dmitriy Dmitriyevich; ROMA-
SENKO, V.A., red.; BEM'CHIKOVA, Yu.S., tekhn.red.

[Textbook on psychiatry] Uchebnik psichistrii. Moskva, Gos.
izd-vo med.lit-ry, 1960. 355 p. (MIRA 13:5)
(PSYCHIATRY)

VERESA, F.I., gornyy inzh.; PORTNOV, A.A., gornyy inzh.; VASHCHENKO, V.S.,
gornyy inzh.

Improving methods of undercutting and blasthole caving. Gor. zhur.
no.6:56-61 Je '63. (MIRA 16:7)

I. Rudnik im. Dzerzhinskogo, Krivoy Rog.
(Mining engineering)

BORINEVICH, V.V.; FEDOTOV, D.D., red.; PORTNOV, A.A., red.

[Narcomanias; clinical aspects, pathogenesis and treatment
of morphinism, codeinism, opiomania, and other opiate
narcomanias] Narkomanii; klinika, patogenez i lechenie mor-
finizma, kodeinizma, opiomani i drugikh opiinykh narkomanii.
Moskva, Gos.nauchno-issl. in-t psichiatrii MZ RSFSR, 1963.
273 p.

(MIRA 16:8)

(NARCOTIC HABIT)

KUDRYAVTSEV, M.A.; PORTNOV, A.A., red.; YAKOVLEVA, N.A., tekhn. red.

[Medical expertise on work capacity in brain injuries] Vrachetno-trudovaya ekspertiza pri travmakh golovnogo mozga. Moskva, Medgiz, 1962. 143 p. (MIRA 16:1)

(BRAIN--WOUNDS AND INJURIES)
(DISABILITY EVALUATION)

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CIA-RDP86-00513R001342520017-6

FEDOTOV, D.D.; PORTNOV, A.A.

Pestintoxicating and postinfectious encephalopathy. Vop. psikh.
no. 4:362-367 '60. (MIRA 15:2)
(BRAIN DISEASES)

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CIA-RDP86-00513R001342520017-6"

PORTNOV, A.A. (Moskva); FEDOTOV, D.D. (Moskva)

Neuroses and psychopathies. Probl.sud.psikh. 9:174-183 '61.
(MIRA 15:2)

(NEUROSES) (MENTAL ILLNESS)

PORNOV, A.A.; BANSHCHIKOV, V.M., prof., red.; FEDOTOV, D.D.,
prof., otv. red.,

[Alcoholism; the characteristics of its development and
course (the clinical aspect of the problem)] Alkogolizm;
osobennosti razvitiia i tcheniia (klinicheskii aspekt
problemy). Pod red. V.M. Banshchikova. Moskva, M-vo zdra-
vookhraneniia RSFSR, 1962. 221 p. (MIRA 15:7)
(ALCOHOLISM)

PORTNOV, A.A. (Krivoy Rog)

Improving the system of working the Dzerzhinskii Mine. Gor.zinur.
no.2:26-29 F '61. (MIRA 14:4)

1. Nahcal'nik tekhnicheskogo otdela rudoupravleniya im. Dzerzhinskogo.

(Krivoy Rog—Mining engineering)

GLINKOV, M.A.; PORTNOV, A.A.

Flame testing of chambers with a vortex flow of gases and a water-cooled receiver in the center near the surface of the brickwork. Izv. vys.ucheb.zav.; chern.met. 4 no.5:184-188 '61. (MIRA 14:6)

1. Moskovskiy institut stali.

(Metallurgical furnaces--Testing)

S/148/61/000/003/014/015
A161/A133

AUTHORS: Glinkov, M. A., Portner, A. A.

TITLE: The gas mechanics in sectional furnaces for rapid steel heating

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy. Chernaya metallurgiya, no. 3, 1961, 172 - 183

TEXT: A detailed description is presented of tests with rapid-heating cyclone type sectional furnaces and those with burners directing hot gas directly on the face of square billets. The flow was studied in furnace models with the aid of water and dye, and with air. Various chamber diameters and different outlet duct dimensions were tested; the test billets were of round and square shape. The article includes diagrams showing the flow speed measured in different spots of the chambers and calculations. The illustrations include a schematic view of the flow in a chamber with burners directed to the center of billet faces. Conclusions: The cyclone motion ensures a dependable ignition and intense burning of gas-air mixture, and the fuel-mixture jets directed to the lining produce a high temperature in the lining and high heat in the chamber space, which in combination results in an intense heat transfer into the billet surface. Convection, causing ✓

Card 1/2

The gas mechanics in sectional furnaces for rapid steel ...A161/A133 S/148/61/000/003/014/015

non-uniform heating in square billets in chambers with large diameter, may be reduced to a minimum by using a larger diameter of the outlet ducts, which lowers the flow speed at the billet and increases the zone of laminar flow. It is not possible to obtain a completely uniform heating of square billets over the entire surface in a cyclone furnace, for the heat transfer from the faces into the center of billet by heat conductivity is faster than that from the ribs. The highest heat supply to surface in the spots of fastest heat transfer from surface to the billet body is achieved when the gas jets are directed to the center of the billet faces, and this arrangement produces more even heating over the entire surface. In cyclone chambers of small diameter the laminar zone at the billet is absent, the gas speed at the billet surface is higher, and the heat transfer by convection more intense. The most advantageous chamber type is of small diameter and with partition walls. There are 6 figures and 4 Soviet-bloc references.

ASSOCIATION: Moskovskiy Institut stali (Moscow Steel Institute)

SUBMITTED: June 21, 1960

Card 2/2

: GLINKOV, M.A.; PORTNOV, A.A.

Gas mechanics in compartment furnaces for the high speed heating
of steel. Izv.vys.ucheb. zav.; chern. met. no.3:172-182 '61.
(MIRA 14:3)

1. Moskovskiy institut stali.
(Furnaces, Heating)
(Gas flow)

SVINARENKO, D.M.; PORTNOV, A.A.

Technical progress and growth of labor productivity at
the "Gigant" Mine. Gor.zhur. no.8:3-7 Ag '60.
(MIRA 13:8)

1. Upravlyayushchiy rudnikom im. Dzerzhinskogo Krivorozhskogo
zhelezorudnogo basseyna (for Svinarenko). 2. Machal'nik
tekhnicheskogo otdela rudoupravleniya im. Dzerzhinskogo
(for Krivorozhskogo zhelezorudnogo basseyna (for Portnov).
(Krivoy Rog--Iron mines and mining)

PORTEOV, A.A.

AVERSHIN, S.G., prof., dokt.tekhn.nauk; ANAN'IN, G.P., dotsent, kand.tekhn.
nauk; BARANOV, A.I., dotsent, inzh.; BERLIN, A.Ye., inzh.;
BOCHKAREV, V.G., kand.tekhn.nauk; BUTKEVICH, R.V., kand.tekhn.nauk;
VESELOVSKIY, V.S., prof., doktor tekhn.nauk; VESKOV, M.I., kand.
tekhn.nauk; VOL'KENAU, A.V., kand.tekhn.nauk; GARKAVI, S.M.,
kand.tekhn.nauk; GORBACHEV, T.F.; DAVIDYANTS, V.T., kand.tekhn.nauk;
DMITRIYEV, M.F., kand.tekhn.nauk; DOBROVOL'SKIY, V.V., kand.tekhn.nauk;
DUKALOV, M.F., kand.tekhn.nauk; ZAITSEV, N.A.; ZARANKIN, P.S., inzh.;
ZVYAGIN, P.Z., dotsent, kand.tekhn.nauk; IL'SHTEYN, A.M., kand.tekhn.
nauk; KILYACHKOV, A.P., dotsent, kand.tekhn.nauk; KIRICHENKO, I.P.,
inzh.; KRUPENNIKOV, G.A., kand.tekhn.nauk; KUZNETSOV, S.T., kand.
tekhn.nauk; KUCHERSKIY, L.V., kand.tekhn.nauk; LINDENAU, N.I., inzh.;
LIPKOVICH, dotsent, kand.tekhn.nauk; LOKSHIN, B.S., kand.tekhn.nauk;
MURATOV, M.L., dotsent, kand.tekhn.nauk; MUCHNIK, V.S., prof.,
doktor tekhn.nauk; NAYDISH, A.M., dotsent, kand.tekhn.nauk; NEKRA-
SOVSKIY, Ya.E., prof., doktor tekhn.nauk; NEKHAYEV, G.A., inzh.;
NUROK, G.A., prof., doktor tekhn.nauk; OVINOV, M.I., inzh.;
PORTEOV, A.A., inzh.; PROSKURIN, V.V., dotsent, kand.tekhn.nauk;
HULNEV, B.A., inzh.; SAPITSKIY, K.F., kand.tekhn.nauk; SELETSKIY, R.A.,
dotsent, kand.tekhn.nauk; SEMENOV, A.P., kand.tekhn.nauk; SKAFI,
P.V., inzh.; SONIN, S.D., prof.; SUDOPLATOV, A.P., prof., doktor
tekhn.nauk; TIMOSHEVICH, V.A., inzh.; FURMAN, A.A., inzh.; CHINAKAL,
N.A.; SHAKHEMEYSTER, L.G., dotsent, kand.tekhn.nauk; TERPIGOREV, A.M.,
glavnnyy red.; LOZNEVA, A.A., red.; NAUMKIN, I.F., red.; OSTROVSKIY,
S.B., red.; PANOV, A.D., red.; STUGAREV, A.S., red.; SHELKOV, A.A.,
(Continued on next card)

AVERSHIN, S.G.---(continued) Card 2.

red.; AUKHANGEL'SKIY, A.S., kand.tekhn.nauk, red.; REZNIKOV, G.A.,
inzh., red.; ALESHIN, M.I., red.izd-va; KACHALKINA, Z.I., red.
izd-va; PROZOROVSKAYA, V.L., tekhn.red.; NADEINSKAYA, A.A., tekhn.red.

[Mining; an encyclopedic handbook] Gornoe delo; entsiklopedicheskii
spravochnik. Glav. red. A.M. Terpigorev. Chleny glav.red.: F.A.
Barabanov i dr. Vol.5 [Underground coal mining] Razrabotka
ugol'nykh mestorozhdenii podzemnym sposobom. Moskva, Gos. nauchno-
tekhn.izd-vo lit-ry po ugol'noi promyshl. 1958. 447 p.

(MIRA 12:2)

1. Chlen-korrespondent Akademii nauk SSSR (for Gorbachev, Chinakal).
2. Chlen-korrespondent Akademii nauk USSR (for Zaytsev).
(Coal mines and mining)

GEYZER, Isay Moiseyevich; PORTNOV, A.A., red.; ZAKHAROVA, A.I., tekhn.red.

[V.V.Veresaev, writer and physician] V.V.Veresaev pisatel'-vrach.
Moskva, Gos.izd-vo med.lit-ry, 1957. 145 p. (MIRA 12:2)
(SMIDOVICH, VIKENTII VIKENT'EVICH, 1867-1945)

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PORTNOV, A. A. Cand Med Sci -- (diss) "Data Concerning the
Problem of the Specificity of Exogenous ~~EXO~~ Psychoses (Acute
Tetraethyl-Lead Psychosis)." Mos, 1957. 12 pp 21 cm. (First
Mos Order of Lenin Medical Inst im I. M. Sechenov), 200 copies
(KL, 18-57, 98)

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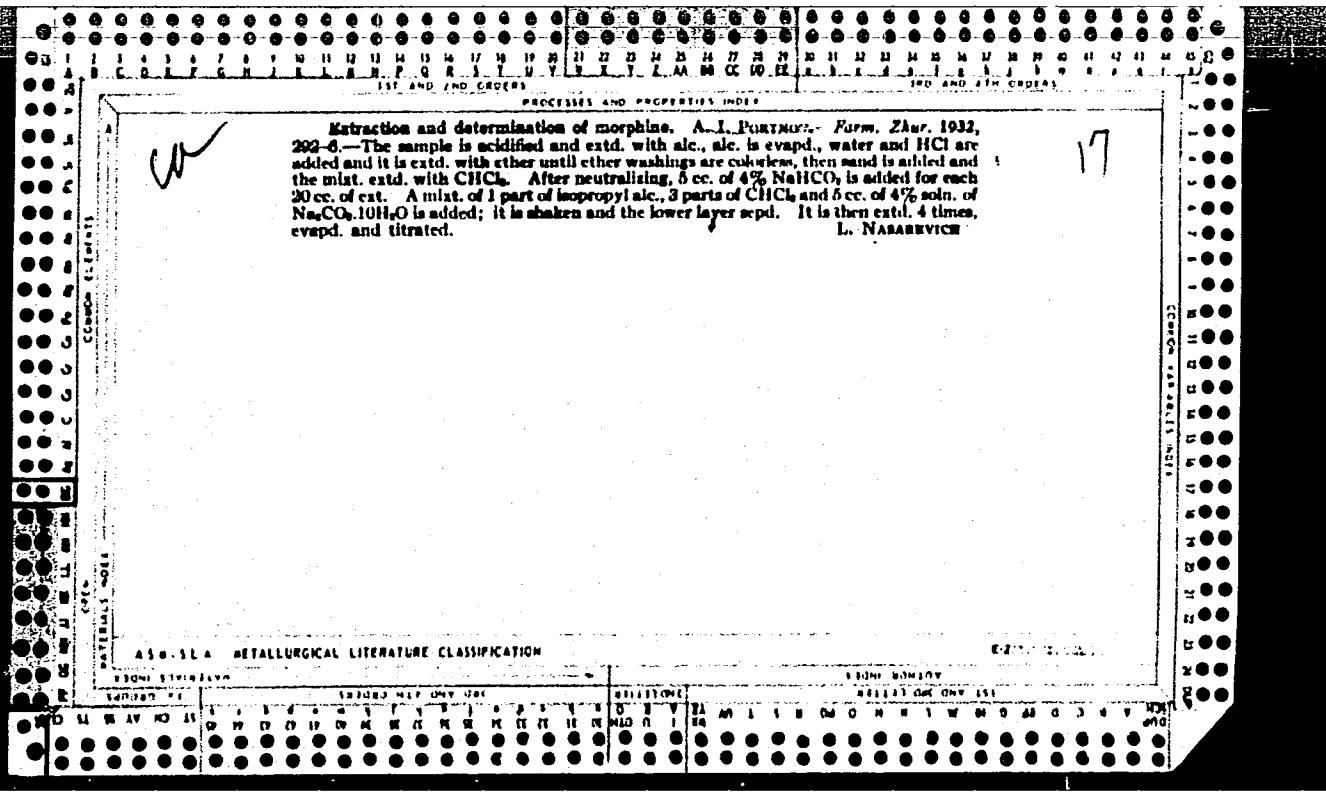
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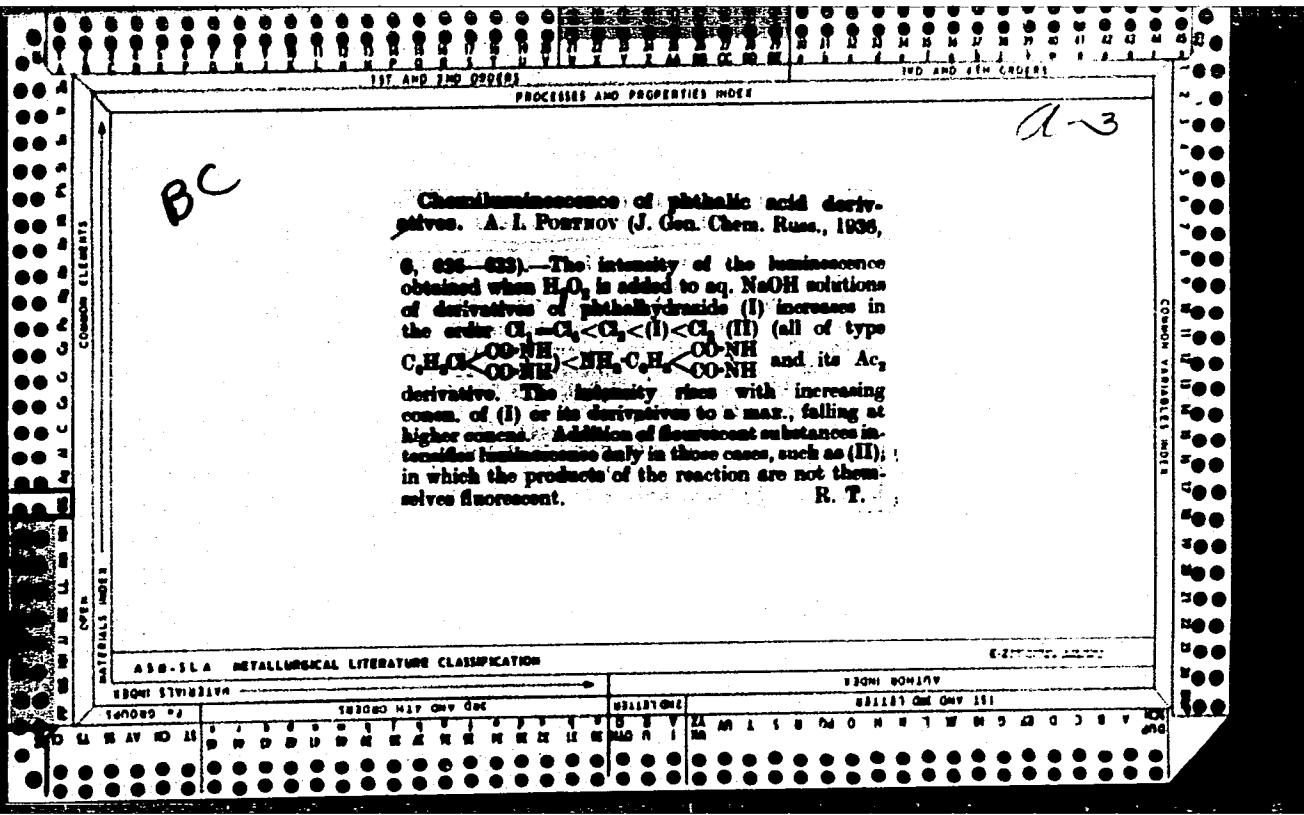
PORNOV, A.A. (Moskva)

Treatment and prophylaxis of psychoneuroses. Med.sestra 18
no.9:3-7 S '59. (MIRA 12:11)
(MENTAL ILLNESS)

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001342520017-6"





Application of aromatic arsenic compounds in chemical analysis. I. A. I. Pastorev. *Zhur. Obrabka. Khim.* (J. Gen. Chem.) 18, 594-600 (1948). -Diwchen. consts. of a no. of aromatic aromatic acids were detd.; at 18° these were: ρ -O₂NC₆H₄AsO₃H, 1.0×10^{-4} ; PhAsO₃H, 2.6×10^{-4} ; ρ -MeC₆H₄AsO₃H, 1.4×10^{-4} ; ρ -HOC₆H₄AsO₃H, 8.2×10^{-4} ; ρ -AcNH₂C₆H₄AsO₃H, 1.5×10^{-4} ; ρ -H₂N-C₆H₄AsO₃H, 4.0×10^{-4} ; the nos. of insol. metal salts found for these acids are: 8, 9, 14, 16, 16, and 18, resp. The latter detns. were made by addn. of 1% soln. of the Na salt of the acid (excess) to a 0.1% soln. of the cation: Cu⁺⁺, Be, Mg, Zn, Al, Th⁺, Mn, Fe⁺⁺, Fe⁺⁺⁺ as sulfates; Ag, Ca, Sr, Cd, Ba, Hg⁺, Bi, and Pb as nitrates; Hg⁺⁺, Se⁺⁺⁺, Sb⁺⁺⁺, Co, and Ni as chlorides; Ta and Hg⁺⁺ as tartrates. The Ph acid gave insol. (or slightly sol. salts) with: Cb, Ag, Sn, Sb, Ta, Hg⁺; the ρ -NH₂ acid gave insol. ppts. with Be, Mg, Al, Mn, Fe, Co, Ni, Cu, Zn, Cd, Ag, Cd, Sn, Sb, Ta, Hg, Pb, Bi; ρ -NO₂ acid was effective with Fe, Cb, Sn, Sb, Ta, Hg⁺, Pb, Bi; ρ -OH acid was effective with Be, Al, Mn, Co, Cu, Zn, Cd, Ag, Cd, Sn, Sb, Ta, Hg, Pb, and Bi; ρ -Me acid gave insol. ppts. with Al, Mn, Fe, Co, Cu, Zn, Cd, Ag, Sn, Sb, Ta, Hg, Pb, Bi; the ρ -AcNH₂ acid gave ppts. with Be, Al, Mn, Fe, Co, Cu, Zn, Cd, Ag, Cd, Sn, Sb, Ta, Hg, Pb, Bi. Study of comparative dissociation of the isomeric NO₂ acids: for σ -NO₂ acid 3.0×10^{-4} , for m -NO₂ acid 7.9×10^{-4} ; the insol. salts were formed by the 1st only with Mn, Co, Fe, Cb, Sn, Ta, Pb, Bi (incomplete with Sb), while the 2nd gave ppts. just like the ρ -isomer. Thus, increased dissociation const. leads to greater solv. of the salts. Interpretation of the results on the basis of electron cloud distribution is made. 18 references.

I. Arsonate method for the determination of cobalt. *Ibid.* 691-4. -The method is based on the different behavior of ρ -AcNH₂C₆H₄AsO₃H₂ to Co and Ni; in the cold only Co is pptd. Metals of the H₂S and (NH₄)₂S groups must be absent; alkali and alk. earth metals do not interfere. The Co salt is apparently a neutral Co salt. The pptn. is slow in the cold (over 10 hrs.) but is very rapid on boiling. The best procedure was as follows: Mix the sample contg. Ni and Co salts with 10 g. pure NH₄Cl and enough H₂O to make the total vol. 50 ml. Heat to boiling and add NH₄OH to make a 0.3 N soln., then a 3-fold excess of the precipitant (5% soln.) with stirring. Cool and again boil 15 min., cool in running water, and let stand 15 min.; wash the filtered ppt. with 33% alc. Dissolve in 6 N HCl at 30-40° and det. the Co in the soln. bromometrically (bromide-bromate method). The presence of Ni causes small pos. errors. **III. Determination of lead in the presence of alkaline earths.** *Ibid.* 605-7. -The method is based on the ability of ρ -HOC₆H₄AsO₃H₂ to ppt. Pb to form a neutral salt. Ba, Ca, and Sr are not pptd. To 10 ml. of Pb^{(NO₃)₂} soln. contg. 0.1-0.15 g. Pb add 25 ml. H₂O and 8 ml. of 5% reagent soln., boil and let stand 1 hr.; wash the filtered ppt. with 33% EtOH. Dissolve it in 25 ml. 2 N H₂SO₄, heating 30 min. on the steam bath. Dil. to 100 ml. and use a 25-ml. aliquot for bromide-bromate titration. One ml. of *N* KBrO₃ is equiv. to 0.2305 g. Pb ρ -hydroxyphenylarsenate. The presence of considerable Ca, Ba, or Sr salts caused only slight errors. G. M. Kosolapoff